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web reaction rates google classroom  $\text{mg s} + 2 \text{hcl a q} \rightarrow \text{mgcl}_2 + 2 \text{a q} + \text{h}_2 \text{g}$  the reaction between  $\text{mg s}$  and  $\text{hcl a q}$  is represented by the equation above in a kinetics experiment a 0.080 g sample of  $\text{mg s}$  ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M  $\text{hcl a q}$  at 25 °C

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reaction reactions conducted in an industrial facility mix compounds together heating and stirring them for a