

INORGANIC CHEMISTRY

451. Nitrogen can be obtained from air by removing
 (a) Oxygen (b) Hydrogen (c) Carbon dioxide (d) Both (a) and (c)
452. Which of the following carbide is a methanide?
 (a) Be_2C (b) CaC_2 (c) Al_2C_3 (d) Mg_2C
453. Caro's acid is
 (a) H_2SO_3 (b) $\text{H}_2\text{S}_2\text{O}_5$ (c) H_2SO_4 (d) $\text{H}_2\text{S}_2\text{O}_8$
454. Inorganic benzene is
 (a) $\text{B}_3\text{H}_3\text{N}_3$ (b) BH_2NH_2 (c) $\text{B}_3\text{H}_6\text{N}_3$ (d) $\text{H}_3\text{B}_3\text{N}_3$
455. For the reaction, $\text{P}(\text{White}) \longrightarrow \text{P}(\text{Red})$; $\Delta H = -18.4 \text{ kJ}$. It follows that
 (a) Red P is readily formed from white P
 (b) White P is readily formed from red P
 (c) White P can not be converted to red P
 (d) White P can be converted into red P and red P is more stable
456. Nitrogen dioxide cannot be obtained by heating
 (a) KNO_3 (b) $\text{Pb}(\text{NO}_3)_2$ (c) $\text{Cu}(\text{NO}_3)_2$ (d) AgNO_3
457. Reactivity of borazole is greater than that of benzene because:
 (a) Borazole is a non-polar compound
 (b) Borazole is a polar compound
 (c) Borazole is an electron deficient compound
 (d) Of localized electrons in it
458. When orthophosphoric acid is heated to 600°C , the product formed is
 (a) Phosphine, PH_3 (b) Phosphorus pentoxide, P_2O_5
 (c) Phosphorus acid, H_3PO_3 (d) Metaphosphoric acid, HPO_3
459. $\text{P}_4 + 3\text{NaOH} + 3\text{H}_2\text{O} \rightarrow \text{A} + 3\text{NaH}_2\text{PO}_2$, here, 'A' is
 (a) NH_3 (b) PH_3 (c) H_2PO_2 (d) H_3PO_4
460. SO_2 in the solid phase exists as
 (a) linear chain or cyclic trimer (b) planar triangular unit
 (c) square planar unit (d) none of these
461. Sulphur does not exist as S_2 molecule because
 (a) It is less electronegative (b) Does not form pπ-pπ bond
 (c) It has ability to exhibiting catenation (d) Has tendency to show variable oxidation state
462. Which of the following has S-S BOND?
 (a) $\text{H}_2\text{S}_2\text{O}_8$ (b) $\text{H}_2\text{S}_2\text{O}_7$ (c) $\text{H}_2\text{S}_2\text{O}_6$ (d) Mustard gas
463. Which of the following is correct order of stability order?
 (a) $\text{Pb}^{2+} > \text{Pb}^{4+}$ (b) $\text{Pb}^{4+} > \text{Pb}^{2+}$ (c) $\text{Si}^{2+} > \text{Si}^{4+}$ (d) $\text{Sn}^{4+} > \text{Sn}^{2+}$
464. Quartz is a crystalline variety of
 (a) Si (b) SiO_2 (c) Na_2SiO_3 (d) SiC
465. The equivalent weight of phosphoric acid in the reaction
 $\text{NaOH} + \text{H}_3\text{PO}_4 \longrightarrow \text{NaH}_2\text{PO}_4 + \text{H}_2\text{O}$
 (a) 25 (b) 49 (c) 59 (d) 98

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 - The Rise of Digital Reading Inorganic Chemistry Questions And Answers Pdf
 - Advantages of eBooks Over Traditional Books
2. Identifying Inorganic Chemistry Questions And Answers Pdf
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Inorganic Chemistry Questions And Answers Pdf
 - User-Friendly Interface
4. Exploring eBook Recommendations from Inorganic Chemistry Questions And Answers Pdf

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 - Inorganic Chemistry Questions And Answers Pdf and Bestseller Lists
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 - Interactive Elements Inorganic Chemistry Questions And Answers Pdf
 8. Staying Engaged with Inorganic Chemistry Questions And Answers Pdf
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Inorganic Chemistry Questions And Answers Pdf
 9. Balancing eBooks and Physical Books Inorganic Chemistry Questions And Answers Pdf
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Inorganic Chemistry Questions And Answers Pdf
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Inorganic Chemistry Questions And Answers Pdf
 - Setting Reading Goals Inorganic Chemistry Questions And Answers Pdf
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Inorganic Chemistry Questions And Answers Pdf

- Fact-Checking eBook Content of Inorganic Chemistry Questions And Answers Pdf
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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