

1. Formaldehyde (CH₂O) is a colorless, pungent gas that is widely used in the chemical industry and as a preservative in formalin solutions.

1.1. Physical Properties

Formaldehyde is a colorless, pungent gas with a boiling point of -19°C and a melting point of -92°C.

1.2. Chemical Properties

Formaldehyde is highly reactive and can undergo various chemical reactions.

1.3. Synthesis

Formaldehyde is synthesized from methanol and carbon monoxide.

1.4. Applications

Formaldehyde is used in the production of plastics, resins, and disinfectants.

1.5. Environmental and Health Concerns

Formaldehyde is a known carcinogen and can cause respiratory irritation.

It is also used in the production of formalin, a preservative for biological specimens.

1.6. Safety Precautions

Formaldehyde should be handled with care and proper ventilation.

1.7. Conclusion

Formaldehyde is a versatile chemical with many uses.

It is important to handle it safely and responsibly.

Formaldehyde is a colorless, pungent gas.

1.8. References

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