

Engineering Thermodynamics

Unit 4 and Unit 5

Question Bank

Part A Questions

1. The percentage of Oxygen present in atmospheric air on mass basis
(a) 21 (b) 23 (c) 79 (d) 77
2. The percentage of Nitrogen present in atmospheric air on mass basis
(a) 21 (b) 23 (c) 79 (d) 77
3. The percentage of Oxygen present in atmospheric air on mass basis
(a) 21 (b) 23 (c) 79 (d) 77
4. The percentage of Nitrogen present in atmospheric air on mass basis
(a) 21 (b) 23 (c) 79 (d) 77
5. The temperature at which water vapour starts condensing is called
(a) Dry bulb temperature (b) Wet bulb temperature
(c) Dew point temperature (d) Boiling temperature
6. Psychrometer measures
(a) Dry bulb temperature (b) Wet bulb temperature
(c) Dew point temperature (d) Both DBT and WBT
7. The X- axis of the psychrometric chart represents
(a) Dry bulb temperature (b) Wet bulb temperature
(c) Specific humidity (d) Relative humidity
8. Which one of the following remains constant during sensible cooling process?
(a) Enthalpy (h) (b) Dry bulb temperature
(c) Specific humidity (W) (d) Relative humidity (ϕ)
9. Which one of the following remains constant during sensible heating process?
(a) Enthalpy (h) (b) Dry bulb temperature
(c) Specific humidity (W) (d) Relative humidity (ϕ)
10. The relative humidity ϕ in percentage for dry air is
(a) 0 (b) 25 (c) 75 (d) 100

Engineering Thermodynamics Question Bank

R. K. Singal, Mridul Singal, Rishi Singal

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Engineering Thermodynamics R. K. Singal, Mridul Singal, Rishi Singal, 2013-12-30 Engineering Thermodynamics has been designed for students of all branches of engineering specially undergraduate students of Mechanical Engineering The book will also serve as reference manual for practising engineers The book has been written in simple language and systematically develops the concepts and principles essential for understanding the subject The text has been supplemented with solved numerical problems illustrations and question banks The present book has been divided in five parts Thermodynamic Laws and Relations Properties of Gases and Vapours Thermodynamics Cycles Heat Transfer and Heat Exchangers Annexures

Guide to RRB Junior Engineer Stage II Mechanical & Allied Engineering 3rd Edition Disha Experts, 2019-03-02 Guide to RRB Junior Engineer Stage II Civil Allied Engineering 3rd Edition covers all the 5 sections including the Technical Ability Section in detail The book covers the complete syllabus as prescribed in the latest notification The book is divided into 5 sections which are further divided into chapters which contains theory explaining the concepts involved followed by Practice Exercises The Technical section is divided into 13 chapters The book provides the Past 2015 2014 Solved questions at the end of each section The book is also very useful for the Section Engineering Exam

Chemical Thermodynamics Siddharth Venkatesh, 2025-02-20 Chemical Thermodynamics The Essentials offers a comprehensive and accessible exploration of the fundamental principles and practical applications of thermodynamics in chemical systems Designed for students researchers and professionals this book delves into the energetic underpinnings of chemical reactions and processes Covering basic principles to advanced topics like phase equilibria and chemical kinetics each chapter provides clear explanations illustrative examples and practical applications The book adopts a rigorous approach to ensure a solid understanding of the subject matter systematically presenting complex concepts and emphasizing a strong theoretical foundation Practical relevance is highlighted through applications in chemical engineering environmental science and materials science Thought provoking exercises accompany each chapter fostering critical thinking and practical problem solving Helpful pedagogical tools such as chapter summaries key terms and glossaries aid comprehension and serve as valuable references Beyond being a textbook Chemical Thermodynamics The Essentials aims to inspire curiosity and exploration in the field of thermodynamics Engaging narratives and insightful discussions encourage readers to delve deeper into the fascinating world of chemical energetics Whether you re a student or a seasoned researcher this book offers a comprehensive and engaging resource to deepen your understanding of chemical thermodynamics and unlock the mysteries of the energetic heart of chemistry

Fundamentals of Chemical Engineering Thermodynamics Themis Matsoukas, 2012-10-02 The Clear Well Organized Introduction to Thermodynamics Theory and Calculations for All Chemical Engineering Undergraduate Students This text is designed to make thermodynamics far easier for undergraduate chemical engineering students to learn and to help them perform thermodynamic calculations with confidence Drawing on his award

winning courses at Penn State Dr Themis Matsoukas focuses on why as well as how He offers extensive imagery to help students conceptualize the equations illuminating thermodynamics with more than 100 figures as well as 190 examples from within and beyond chemical engineering Part I clearly introduces the laws of thermodynamics with applications to pure fluids Part II extends thermodynamics to mixtures emphasizing phase and chemical equilibrium Throughout Matsoukas focuses on topics that link tightly to other key areas of undergraduate chemical engineering including separations reactions and capstone design More than 300 end of chapter problems range from basic calculations to realistic environmental applications these can be solved with any leading mathematical software Coverage includes Pure fluids PVT behavior and basic calculations of enthalpy and entropy Fundamental relationships and the calculation of properties from equations of state Thermodynamic analysis of chemical processes Phase diagrams of binary and simple ternary systems Thermodynamics of mixtures using equations of state Ideal and nonideal solutions Partial miscibility solubility of gases and solids osmotic processes Reaction equilibrium with applications to single and multiphase reactions

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 Narinder Kumar, Dr. J. K. Juneja, 2006 *Engineering*, 1925 *Engineering Magazine*, 1912 *Engineering Record*,
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