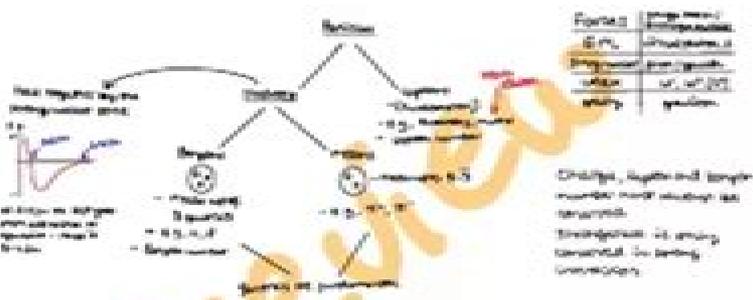


Particles and Quantum



To convert mass to energy, use $E=mc^2$
 Annihilation - pair of particles and its corresponding antiparticle collide and their rest energy is converted into photons

Pair production - when photon turns into a particle and its corresponding antiparticle. The photon must have at least the sum amount of energy which was energy of the particles

Excitation
 Existing atoms they gain energy through energy to climb their atomic levels
 Alpha particles - Helium nucleus, ^4_2He

- highly ionizing, as it is very heavy
 - relatively penetrating - stopped by a plate of paper or foil on their

Beta - fast moving electron $\rightarrow e^-$
 - medium penetrating and velocity varying - stopped by a few cm of aluminium
 - fast moving - $\frac{1}{2}c$ to $\frac{3}{4}c$ - $\frac{1}{2}c$ - penetrates to several centimeters



Gamma - high energy EM ray or photon emitted from a high energy nucleus $\rightarrow \gamma$
 - velocity varying
 - highly penetrating - stopped by a few m of concrete

Isotopes - same element - same number of protons
 - different relative atomic mass - different number of neutrons
 $^{12}_6\text{C}$ and $^{14}_6\text{C}$

Nuclear charge - $\frac{Q}{4\pi\epsilon_0} \left(\frac{1}{r^2} \right)$
 - charge to mass ratio

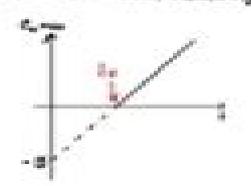
Radioactive - change of a nucleus when spontaneous through a bit of time and
 - $^{238}_{92}\text{U}$
 - $^{14}_6\text{C}$



Photoelectric effect

Use photons of sufficient energy on electrons on the surface of the metal, liberating them

- $E_{\text{photon}} = hf = \phi + E_{\text{kin}}$
- ϕ - photon absorbed - $h\nu$ must be $\geq \phi$ (work function)
- E_{kin} - kinetic energy
- ϕ - work function - minimum energy to liberate electrons
- $E_{\text{photon}} = hf$ - energy that will give an e^- is conserved
- E_{kin} - stopped potential - minimum potential needed for electrons to be liberated



The photoelectric effect proved the particle nature of light due to the fact that the maximum kinetic energy of electrons E_{kin} is not linearly dependent on the wavelength of light. Instead, it is linearly dependent on the frequency of light. The work function ϕ is the minimum energy needed for an electron to be liberated from the metal surface.

Stopping potential - the pot. needed to stop any electrons from reaching the other side

- $E_{\text{photon}} = hf$
- ϕ - work function
- E_{kin} - stopping potential

De Broglie

The de Broglie wavelength is the wavelength that a particle can have. The wave nature of a particle was proven by diffraction of a crystal. This is because the electron is diffracting which is wave nature.



- $\lambda = \frac{h}{p}$
- $\lambda = \frac{h}{mv}$
- λ - de Broglie wavelength
- p - momentum / mass x velocity

Fluorescent tubes

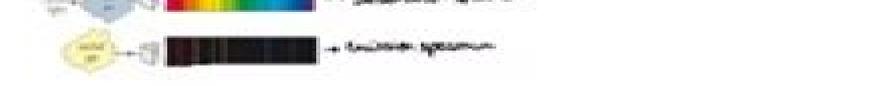


These tubes contain mercury under a high voltage is applied. This excites the gas molecules to which some of the mercury atoms emit photons when they relax. When the gas relaxes it emits light. The light is higher energy than when the electrons relax, but the photons in the tube are of a longer wavelength. It then causes the phosphor coating to emit visible light.

Energy levels

Electrons can be excited to higher energy levels. Either by absorbing photons from the light or by the collision. In a gas discharge tube, when an electron collides with a gas atom, it can transfer its energy to the atom.

Ionization energy - is enough energy to give an electron, so will completely remove the electron



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