

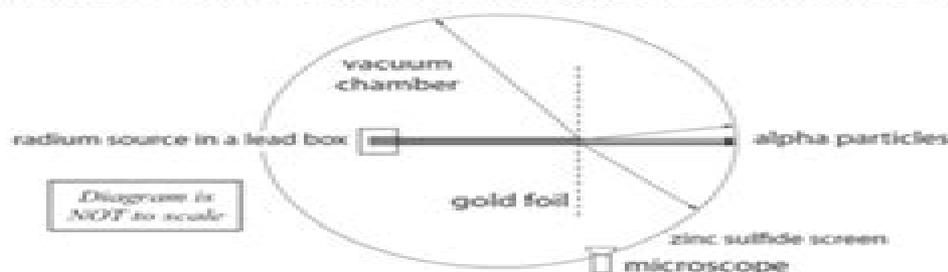
## ATOMS: ATOMIC STRUCTURE QUESTIONS AND ANSWERS

### QUESTION ONE: MODELS OF THE ATOM (2011:1)

At different times scientists have proposed various descriptions or models of the atom to match experimental evidence available.

- (a) The model that Thomson proposed was called the plum-pudding model. Describe this model.  
**Thomson's model had a number of negative particles / corpuscles / electrons embedded in a (solid) positive sphere / cloud.**  
**Cloud of particles or energy not accepted.**  
**The atom as a whole is neutral because the charges are balanced.**  
**Term "nucleus" not accepted.**

Geiger and Marsden performed a series of experiments under the direction of Ernest Rutherford which led to a new model of the atom. A model of the gold foil experiment is shown below.



- (b) For each of the conclusions given below, state which observation from the experiment provides evidence that:
- (i) Most of the mass of the atom is concentrated in a tiny region which Rutherford called the nucleus.  
**Most of the alpha particles went straight through and a very small number / a few bounced back / were reflected.**
- (ii) The nucleus is positively charged.  
**Some of the alpha particles were deflected ("deflection" is an observation, "repulsion" is not).**
- (c) Rutherford included several features in the experiment due to the characteristics of the alpha particles. Discuss the reason for each of the features below:
- (i) The source of the alpha particles is at the end of a narrow lead tube  
**Tube: To provide a narrow (or collimated) beam of alpha particle.**  
**OR to direct the alpha particles**  
**OR ensure alpha hits perpendicular to gold foil**  
**OR to avoid stray alpha particles affecting the results**
- (ii) Vacuum chamber  
**Vacuum chamber – Alpha particles would not travel far / be stopped / lose energy / path would change / be deviated / collide with the air particles (cannot accept "affect", "disrupt", "react", "interfere").**  
**OR**  
**Alpha would become a helium atom / alpha particles ionise the air.**
- (iii) Zinc sulfide screen  
**Alpha particles are not visible to the naked eye. The ZnS coating will show / glow / fluoresce / flash / give off light when alpha particles strike it (this may be implied as opposed to directly stated) but do not allow "detected" or "located"**

# Atoms Atomic Structure Questions And Answers

**A Gutmann**



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