

Solutions for Gravimetric Analysis Questions

Check for Understanding 4.1

1. Determine the solubility of AgCl using K_{sp} for AgCl and a table of initial and equilibrium concentration terms.

	$\text{AgCl}(s)$	\rightleftharpoons	$\text{Ag}^+(\text{aq})$	$+$	$\text{Cl}^-(\text{aq})$	
init			0.0060		0	
equil			$0.0060 + x$		x	where x = increase in $[\text{Cl}^-]$ = solubility of AgCl

at equil, $K_{sp} = 1.8 \times 10^{-10} = [\text{Ag}^+][\text{Cl}^-] = (0.0060 + x)(x) = 0.0060x \rightarrow x = 3.0_0 \times 10^{-8} \text{ M}$

2. $200. \text{ mL} \times \frac{3.0_0 \times 10^{-8} \text{ mol AgCl}}{1000 \text{ mL}} \times \frac{143.32 \text{ g AgCl}}{\text{mol AgCl}} = 8.6 \times 10^{-7} \text{ g AgCl}$

Check for Understanding 4.2

1. Often the solubility of a slightly soluble salt is increased in acidic solution because the anion combines with H^+ to form a weak acid. Removal of the anion shifts the equilibrium toward more dissolution of the salt. Since Cl would form a strong acid (HCl) that is ~100% ionized in solution, it is not removed from solution. Therefore, AgCl is not more soluble in acidic solution and the presence of HNO_3 does not increase its solubility.

Check for Understanding 4.3

1. First, determine the solubility of AgCl using K_{sp} for AgCl and a table of initial and equilibrium concentration terms.

	$\text{AgCl}(s)$	\rightleftharpoons	$\text{Ag}^+(\text{aq})$	$+$	$\text{Cl}^-(\text{aq})$	
init			0		0	
equil			x		x	

Gravimetric Analysis Calculation Questions

IM Harris



Gravimetric Analysis Calculation Questions:

Unveiling the Magic of Words: A Overview of "**Gravimetric Analysis Calculation Questions**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Gravimetric Analysis Calculation Questions**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect on the souls of its readers.

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