

# MOLE CONCEPT

## SHORTNOTE

### ★ LAW OF CHEMICAL COMBINATIONS

- ① LAW OF CONSERVATION OF MASS =  
Total mass of Reactant = Total mass of product
- ② GAZ VUSSAC'S LAW = Gases combine or are produced in a chemical reaction in the simple ratio by volume provided all gases at the same temperature and pressure.
- ③ CONSTANT COMPOSITION = A given compound always contains exactly the same proportion of element by weight.
- ④ RECIPROCAL PROPORTION = When two different elements combined with a fixed weight of third element, the ratio of their combination will either be same or simple multiple of the ratio of which they combine with each other.
- ⑤ MULTIPLE PROPORTION = Atoms of two or more elements may combine in different ratio to produce more than one compound.

# Mole Concept Theory Notes Slibforme

**M Tight**



## **Mole Concept Theory Notes Slibforme:**

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