

**CHAPTER 6 REVIEW***Chemical Bonding***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

\_\_\_\_\_ a. ionic bond  
 \_\_\_\_\_ b. covalent bond  
 \_\_\_\_\_ c. metallic bond

2. Use the electronegativity values shown in Figure 5-20, on page 151 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ a. H—F	_____ d. H—H
_____ b. Na—Cl	_____ e. H—C
_____ c. H—O	_____ f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

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4.  $\text{H}_2\text{S}$  and  $\text{H}_2\text{O}$  have similar structures and their central atoms belong to the same group. Yet  $\text{H}_2\text{S}$  is a gas at room temperature and  $\text{H}_2\text{O}$  is a liquid. Use bonding principles to explain why this is true.

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# Holt Modern Chemistry Chapter 6 Review Packet Answers

**S Ben Porath**



**Holt Modern Chemistry Chapter 6 Review Packet Answers:**

## Decoding **Holt Modern Chemistry Chapter 6 Review Packet Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Holt Modern Chemistry Chapter 6 Review Packet Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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