



**Across**

- 3. A chemical equation that has the same amount of each element on both sides of the equation. **Balanced Equation**
- 4. The amount of mass that is in the reactions should match the amount of mass in the products. **Conservation of Mass**
- 6. The relationship between the amount of moles in a chemical reaction. **Mole Ratio**
- 8a. The mass of a single element or compound in one mole. **Molar Mass**
- 8b. A basic unit of a chemical element **Atom**
- 15. A way to changing from one set of units to another. **Conversion Factor**
- 17. The relationship between the amount of substances in a chemical reaction.

State Chemistry

**Down**

- 1. The amount of molecules in one mole. **Avogadro's Number**
- 2. The number in front of a chemical equation, used to balance. **Coefficient**
- 5. How we measure the mass of a molecule. **Grams**
- 6. Equal to  $6.02 \times 10^{23}$  **Mole**
- 7. The chemical formula for an ionic compound **Formula Unit**
- 9. The amount of a particular elements or molecules in a chemical equation. **Subscript**
- 11. The mass of all the protons, neutrons, and electrons in an element. **Atomic Mass**
- 12. The amount of moles in 120 g of water **Seven**
- 13. A group of two or more atoms **Molecule**
- 16. The amount of moles in 32 g of oxygen **Two**

# Stoichiometry Crossword Puzzle Answers

**Karin Nielsen-Saines**



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