

STEVEN JACOBS

PROFESSIONAL TITLE

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SUMMARY

Your summary goes here. It's the **last stage in the resume writing process**, so pick it up in the end. Focus more on what value you can provide, instead of what you are looking for. Limit it to **2-3 lines max**.

KEY SKILLS

Market Research
Social Media Strategy
Social Media Analytics • Social Media Content Creation & Curation
Marketing Collateral
Keyword Research
Search Engine Optimization
Client Servicing
Blog Writing
Digital Marketing
Website Content Proofreading
Content Management & Editing

TECHNICAL SKILLS

Languages: VB6, SQL, Web Services - C#, C, C++, Java, Perl

- **Operating Systems** : Windows (XP, NT 3.x- 4.0, 2000, 2003) and UNIX
- **Databases**: MS SQL Server (2008, 2005, 2000, 7.0, 6.5, 6.0), MS Access, FoxPro, DTS, DB2
- **Reporting**: Crystal Report, SQL Server 2005 Reporting Services (SSRS)

INTERNSHIPS

Enter official designation as per company records Sep '18 - Nov '18
Enter company name | Enter location (with state -)
Enter a one-line description of the company where you were working, with figures around revenue, geographic presence, no. of clients, etc.
Some rules will apply as discussed above

PROFESSIONAL EXPERIENCE

Enter the official designation as per company records Jun '19 - Sep '19
Enter company name in full | Enter location (with state -)
Enter a one-line description of the company where you were working, with figures around revenue, geographic presence, no. of clients, etc.

- Use one-liner points to present the details of your internship/part-time/professional experience
- Rename the section heading as *Relevant Experience* (if you want to show all your roles - internships, part-time, volunteering, etc. - in one section).
- **Bold keywords and phrases** where you would like to draw the recruiter's attention.
- Use **action verbs** to start off your points, and **achievement figures** (ballpark/approx figures will work as well) to demonstrate the extent of your contributions
- **It's the first step in the resume writing process.** Start off with updating this section first. Based on the keywords and functions you bold, accordingly update the Skills section. Based on that, update your summary. Make sure only those functions are mentioned in the Summary/Skills which have been substantiated in this section.
- Instead of writing all your one-liner points together, group similar points under unique subheadings as shown below:

Social Media Marketing

- Conducted research on **industry trends** & wrote compelling content for **social media posts**; created **5+ posts** daily
- Tracked **social media analytics** via HootSuite & other tools, and reported **results & new ideas** to our marketing team
- Created content to be shared by **20+ influencers** on Instagram, Facebook, and blog accounts to increase **traffic by 33%**

Content Management

- Liaised with the Editorial **team of 25** to manage social media content & community, while coordinating with **30+ freelancers**
 - Increased Instagram & Facebook **followers by 20% and 25%** respectively in **3 months**
- Coordinated with the Design team to create platform-specific social media assets such as **flyers, ads, videos, GIFs**, etc.
- Received **2 million+ likes** on **20+ posts** on Facebook, Twitter, and Instagram

EDUCATION

Enter your official degree (with major) Jun '12 - May '15

Resume Template Ideas

Evelyn U. Salvador



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Adopting the Melody of Appearance: An Emotional Symphony within **Resume Template Ideas**

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this video accompanies the chm 252 lab nitration of bromobenzene in this video students will get an introduction to the lab as well as the set up for conducting the reaction students should

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and

explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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lab report 15 nitration of bromobenzene brynn abbott amp katie muerer bea0006 uah ch 336 02 september 21 2022 abstract in this experiment the electrophilic aromatic substitution mechanism was used to obtain 4-nitrobromobenzene and 2-

nitrobromobenzene recrystallization was used to purify 4-nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o-bromonitrobenzene

and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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