

Alka Seltzer Reactions: An Inquiry-Based Investigation

In your lab groups, you will design, execute and interpret an experiment to test the effect of particle size on the reaction between Alka Seltzer tablets and water. You should be able to quantify your results in *at least* 1 way. Stronger experiments will include more than 1 type of quantitative data. (*Remember: "Quantitative" means you will be taking a measurement of something)

BACKGROUND:

Alka Seltzer is a combination of an acid ($C_6H_8O_7$) and baking soda ($NaHCO_3$). When the two react in water, they generate, among other things, Carbon Dioxide gas (CO_2).



This means that as the reaction proceeds, you will observe several changes, including bubbles that indicate the release and loss of the carbon dioxide gas. You must use these changes in order to quantify the rate and extent of the reaction.

PROBLEM: How will the size of Alka Seltzer particles affect the rate and extent of its reaction with water?

PREDICTION: What result do you expect to see in this experiment and why?

MATERIALS AVAILABLE:

- | | |
|-----------------------|------------------------|
| - Alka Seltzer | - Balloons |
| - Water | - Beakers/cups |
| - Graduated Cylinders | - Plastic soda bottles |
| - Stop Watch or Timer | - Foil |
| - Balance | - Syran wrap |
| - Straws | - Rubber bands |

DIRECTIONS:

1. Discuss with your group the goal of this lab.
2. Examine the available materials and write a procedure for your experiment.
3. Before starting experimentation, check in with your teacher and have your procedures approved.
4. Conduct your experiment and make adjustments as needed- you may need to conduct more than one trial or try several methods in order to produce quality data.
5. Be sure to include all procedures with your lab report.

Alka Seltzer Lab Answers

Sebastian Brünink



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