

NB: Iron cross membrane as:

- Ferrous state Fe^{+2} .
- Ferric state Fe^{+3} conjugated with protein.

▣ occurs at → duodenum & proximal part of jejunum.

▣ absorption is enhanced → When iron stores in body are depleted.

NB: When adequate quantity of iron is stored, absorption is ↓.

▣ **Only ferrous** (and not ferric) form of iron **is absorbed**.

▣ Ferric iron is reduced to ferrous iron by ferric reductase (an enzyme present on surface of enterocytes).

▣ Ferrous iron in intestinal lumen binds to mucosal cell protein, called **divalent metal transporter-1 (DMT-1)**.

▣ This bound iron is then transported into mucosal cell.

▣ rest of the unabsorbed iron is excreted.

▣ **Inside intestinal cell:**

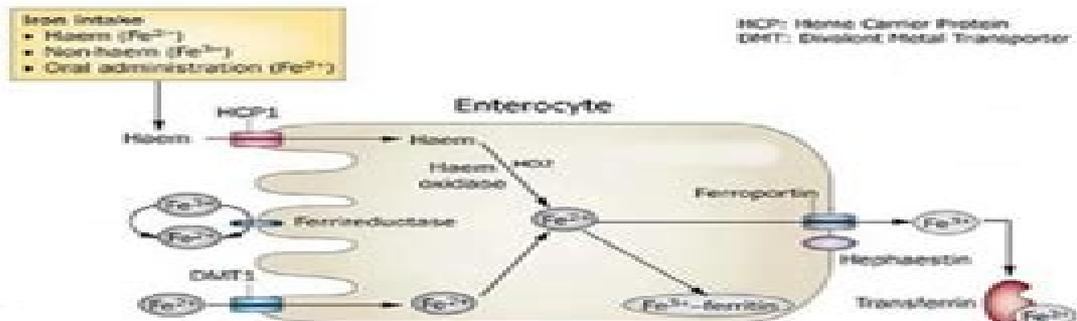
- iron is oxidized to ferric state.
- and **combined with apoferritin** to form **ferritin** (main storage form in intestinal cells). kept temporarily in the mucosal cell.

NB: in case of anemia, iron is further absorbed into bloodstream.

▣ If transferrin is saturated with iron, any iron accumulated in mucosal cell is lost **when cell is desquamated**.

▣ fraction of iron absorbed & retained is decided by iron status.

▣ When iron is in excess, absorption is reduced; this is basis of "mucosal block".



Biochemistry Easy

Michael Seilmaier



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Table of Contents Biochemistry Easy

1. Understanding the eBook Biochemistry Easy
 - The Rise of Digital Reading Biochemistry Easy
 - Advantages of eBooks Over Traditional Books
2. Identifying Biochemistry Easy
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biochemistry Easy
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biochemistry Easy
 - Personalized Recommendations
 - Biochemistry Easy User Reviews and Ratings
 - Biochemistry Easy and Bestseller Lists
5. Accessing Biochemistry Easy Free and Paid eBooks
 - Biochemistry Easy Public Domain eBooks
 - Biochemistry Easy eBook Subscription Services
 - Biochemistry Easy Budget-Friendly Options
6. Navigating Biochemistry Easy eBook Formats

- ePub, PDF, MOBI, and More
 - Biochemistry Easy Compatibility with Devices
 - Biochemistry Easy Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biochemistry Easy
 - Highlighting and Note-Taking Biochemistry Easy
 - Interactive Elements Biochemistry Easy
 8. Staying Engaged with Biochemistry Easy
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biochemistry Easy
 9. Balancing eBooks and Physical Books Biochemistry Easy
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biochemistry Easy
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Biochemistry Easy
 - Setting Reading Goals Biochemistry Easy
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Biochemistry Easy
 - Fact-Checking eBook Content of Biochemistry Easy
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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