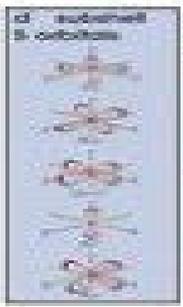


CHEMISTRY

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PERIODIC TABLE OF THE ELEMENTS

Labels for the periodic table:
 H — atomic number
 1.008 — atomic weight
 H — atomic symbol



ATOMIC STRUCTURE

Atomic Number, Z , of an element is the number of protons in the nucleus. For neutral atoms, $Z = +$ the number of electrons. The mass number, A , is the sum of the number of protons and the number of neutrons. Isotopes of an element have the same Z , different A . The atomic weight is the weighted average of the atomic masses of the naturally occurring isotopes of an element.

ATOMIC QUANTUM NUMBERS AND ORBITALS

Principal quantum number, n , is the number of nodes in the radial wave function. It is the number of shells. $n = 1, 2, 3, \dots$. The number of orbitals in a shell is n^2 . The number of orbitals in a subshell is $2l + 1$. The number of orbitals in a subshell is $2l + 1$. The number of orbitals in a subshell is $2l + 1$.

NUCLEAR CHEMISTRY

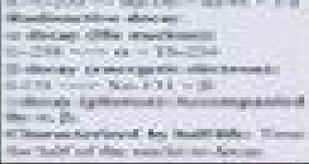
Nuclear reactions occur when nucleons (protons and neutrons) are rearranged. The energy released or absorbed is given by the mass defect. The mass defect is the difference between the mass of the reactants and the mass of the products. The energy released is given by $E = mc^2$.

MANY-ELECTRON ATOMS

Orbitals and quantum numbers are used to describe the electronic structure of atoms. The energy levels of many-electron atoms are split into subshells. The energy levels of many-electron atoms are split into subshells. The energy levels of many-electron atoms are split into subshells.

n	l	m _l	orbitals
1	0	0	1 orbital
2	0	0	1 orbital
2	1	0, ±1	3 orbitals
3	0	0	1 orbital
3	1	0, ±1	3 orbitals
3	2	0, ±1, ±2	5 orbitals
4	0	0	1 orbital
4	1	0, ±1	3 orbitals
4	2	0, ±1, ±2	5 orbitals
4	3	0, ±1, ±2, ±3	7 orbitals

Atomic configuration: The arrangement of electrons in an atom is described by the atomic configuration. The atomic configuration is the arrangement of electrons in an atom. The atomic configuration is the arrangement of electrons in an atom.



Block	Group	Element	Symbol	Atomic Number		
s-block	1	H	H	1		
		Li	Li	3		
	2	Na	Na	11		
		K	K	19		
	p-block	13	B	B	5	
			Al	Al	13	
		14	C	C	6	
			Si	Si	14	
		15	N	N	7	
			P	P	15	
		16	O	O	8	
			S	S	16	
		17	F	F	9	
			Cl	Cl	17	
		18	Ne	Ne	10	
			Ar	Ar	18	
		d-block	3	Sc	Sc	21
				Ti	Ti	22
4	V		V	23		
	Cr		Cr	24		
5	Mn		Mn	25		
	Fe		Fe	26		
6	Co		Co	27		
	Ni		Ni	28		
7	Cu		Cu	29		
	Zn		Zn	30		
f-block	7	La	La	57		
		Ce	Ce	58		
	8	Pr	Pr	59		
		Nd	Nd	60		
	9	Pm	Pm	61		
		Sm	Sm	62		
	10	Eu	Eu	63		
		Gd	Gd	64		
	11	Tb	Tb	65		
		Dy	Dy	66		
12	Ho	Ho	67			
	Er	Er	68			
13	Tm	Tm	69			
	Yb	Yb	70			
14	Lu	Lu	71			
	Hf	Hf	72			

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