

## Test "Chemistry"

## Chapter # 3

### Objective part

Time: 15 min

Total marks: 12

Q.1- Every question has four possible answers. Encircle the correct option.

- i) **The atomic radii of the elements in periodic table;**  
a) increase from left to right in a period  
b) increase from top to bottom in a group  
c) do not change from left to right in a period  
d) decrease from top to bottom in a group
- ii) **Mendeleev Periodic table was based upon the:**  
a) electronic configuration  
b) atomic mass  
c) atomic number  
d) electro negativity
- iii) **The amount of energy given out when an electron is added to an atom is called;**  
a) ionization energy  
b) lattice energy  
c) electron affinity  
d) electro negativity
- iv) **Long form of periodic table is constructed on the basis of:**  
a) Mendeleev postulate  
b) atomic number  
c) atomic mass  
d) mass number
- v) **Along the period which of the following decreases:**  
a) electro negativity  
b) ionization energy  
c) atomic radius  
d) electron affinity
- vi) **Mark the incorrect statement about the ionization energy:**  
a) it is measured  $\text{kJ mol}^{-1}$   
b) it is the absorption of energy  
c) it decreases in a period  
d) it decreases in groups
- vii) **Transition elements are:**  
a) all gases  
b) all metals  
c) all non metals  
d) all metalloids
- viii) **S-f series of inner transition elements are called:**  
a) lanthanides  
b) actinides  
c) halogens  
d) alkali metals
- ix) **If we move from left to right in a period the value of ionization energy:**  
a) remain same  
b) decreases  
c) increases  
d) not effected
- x) **Which scientist has given the idea of actives in periodic table:**  
a) Mendeleev  
b) Al-razi  
c) Newland  
d) dobernier
- xi) **Shielding effect is greater in atoms with greater number of:**  
a) proton  
b) electron  
c) neutron  
d) positron
- xii) **Which is the incomplete period in the periodic table:**  
a) 4th period  
b) 2nd period  
c) 7th period  
d) 1st period

# Chemistry Chapter 3 Test Answers

**J Dewey**



## Chemistry Chapter 3 Test Answers:

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