

Chapter 2 - Practice Test #1

7. An equilibrium system shifts left when the
- rate of the forward reaction is equal to the rate of the reverse reaction.
 - rate of the forward reaction is less than the rate of the reverse reaction.
 - rate of the forward reaction is greater than the rate of the reverse reaction.
 - rate of the forward reaction and the reverse reaction are constant.
8. A 1.00 L flask contains a gaseous equilibrium system. The addition of reactants to this flask results in a
- shift left and a decrease in the concentration of the products.
 - shift left and a increase in the concentration of the products.
 - shift right and a decrease in the concentration of the products.
 - shift right and a increase in the concentration of the products.
9. Consider the following equilibrium: $\text{CH}_4(g) + \text{H}_2\text{O}(g) + \text{heat} = \text{CO}(g) + 3\text{H}_2(g)$
In which of the following will both stresses shift the equilibrium to the right?
- a decrease in temperature and a decrease in volume
 - a increase in temperature and a decrease in volume
 - a decrease in temperature and a increase in volume
 - a increase in temperature and a increase in volume
10. Consider the following equilibrium: $2\text{SO}_2(g) + \text{O}_2(g) = 2\text{SO}_3(g)$ $\Delta H = -198 \text{ kJ}$
There will be no shift in this equilibrium when
- more O_2 is added.
 - a catalyst is added.
 - the volume is increased.
 - the temperature is increased.
11. Consider the following equilibrium: $2\text{Fe}_2\text{O}_3 + 3\text{H}_2\text{O}(g) = \text{Fe}_2\text{O}_3 + 3\text{H}_2$
The equilibrium expression is
- $$\text{Keq} = \frac{[\text{Fe}_2\text{O}_3][\text{H}_2]^3}{[\text{Fe}]^2[\text{H}_2\text{O}]^3}$$
 - $$\text{Keq} = \frac{[\text{Fe}_2\text{O}_3][3\text{H}_2]}{[2\text{Fe}][3\text{H}_2\text{O}]}$$
 - $$\text{Keq} = \frac{[\text{H}_2]^3}{[\text{H}_2\text{O}]^3}$$
 - $$\text{Keq} = [\text{H}_2]^3$$
12. Consider the following equilibrium: $\text{N}_2\text{O}_4 = 2\text{NO}_2$ $\text{Keq} = 0.133$
At equilibrium, the $[\text{N}_2\text{O}_4]$ is equal to

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web processes the type of wastewater treatment used by the facility might affect the frequency of effluent monitoring an industrial facility employing biological treatment would have a similar monitoring frequency as a secondary treatment plant with the same units used for wastewater treatment

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web the dally flow 8 liters x 3 24 liters 30 llters x 1 30 liters the dally flow 54 liters per day on site estimating this method is similar to on site measurement except that the following tables are used to estimate quantities instead of measuring the capacity of

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web purpose this document describes general and specific procedures methods and considerations to be used and observed when conducting wastewater flow measurement scope application the procedures contained in this document are to be used by field personnel when conducting wastewater flow measurement

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