

Chemical Engineering Basics MCQ Practice Test - Set 01

MCQ Chemical Basics

Practice Test: Question Set – 01

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1. Metal cutting by cryo-welding flame is accomplished by the _____ of the metal.

- (A) Shrinking
 - (B) Intensive oxidation
 - (C) Evaporation
 - (D) None of these
-

2. The only suitable method for hardening the low carbon steel is case hardening. Which of the following is a case hardening process?

- (A) Cyaniding
 - (B) Sulfurizing
 - (C) Sybernizing
 - (D) None of these
-

3. Filler material used in welding should have _____, as compared to the parent metal to be welded.

- (A) Lower melting temperature
- (B) Same melting temperature
- (C) Same composition

Chemical Engineering Test Questions Answers

Wolfgang Guggemos



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Metallurgical & Chemical Engineering Eugene Franz Roeber, Howard Coon Parmelee, 1917 [Basic Principles and Calculations in Chemical Engineering](#) David M. Himmelblau, James B. Riggs, 2022-07-27 The 1 Guide to Chemical Engineering Principles Techniques Calculations and Applications Revised Streamlined and Modernized with New Examples Basic Principles and Calculations in Chemical Engineering Ninth Edition has been thoroughly revised streamlined and updated to reflect sweeping changes in the chemical engineering field This introductory guide addresses the full scope of contemporary chemical petroleum and environmental engineering applications and contains extensive new coverage and examples related to biotech nanotech green environmental engineering and process safety with many new MATLAB and Python problems throughout Authors David M Himmelblau and James B Riggs offer a strong foundation of skills and knowledge for successful study and practice guiding students through formulating and solving material and energy balance problems as well as describing gases liquids and vapors Throughout they introduce efficient consistent learner friendly ways to solve problems analyze data and gain a conceptual application based understanding of modern processes This edition condenses coverage from previous editions to serve today s students and faculty more efficiently In two entirely new chapters the authors provide a comprehensive introduction to dynamic material and energy balances as well as psychrometric charts Modular chapters designed to support introductory courses of any length Introductions to unit conversions basis selection and process measurements Strategies for solving diverse material and energy balance problems including material balances with chemical reaction and for multi unit processes and energy balances with reaction Clear introductions to key concepts ranging from stoichiometry to enthalpy Coverage of ideal real gases multi phase equilibria unsteady state material humidity psychrometric charts and more Self assessment questions to help readers identify areas they don t fully understand Thought discussion and homework problems in every chapter New biotech bioengineering nanotechnology green environmental engineering and process safety coverage Relevant new MATLAB and Python homework problems and projects Extensive tables charts and glossaries in each chapter Reference appendices presenting atomic weights and numbers Pitzer Z 0 Z 1 factors heats of formation and combustion and more Easier than ever to use this book is the definitive practical introduction for students license candidates practicing engineers and scientists Supplemental Online Content available with book registration Three additional chapters on Heats of Solution and Mixing Liquids and Gases in Equilibrium with Solids and Solving Material and Energy Balances with Process Simulators Flowsheeting Codes Nine additional appendices Physical Properties of Various Organic and Inorganic Substances Heat Capacity Equations Vapor Pressures Heats of Solution and Dilution Enthalpy Concentration Data Thermodynamic Charts Physical Properties of Petroleum Fractions Solution of Sets of Equations Fitting Functions to Data Register your book for convenient access to downloads updates and or corrections as they become available See inside book for details [Basic Principles and Calculations in Chemical Engineering](#) David

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