

**CLASS VIII**  
**SUBJECT: CHEMISTRY**  
**CHAPTER: 6 – 'IONIC BONDING' & CHAPTER: 7 – 'COVALENT & METALLIC BONDING'**  
**PRACTICE QUESTION-ANSWER**

**QUESTION # 1**

The following diagram shows the structural formula of four covalent compounds:



Draw dot and cross diagram for each of the molecules.

**ANSWER:**

Dot and cross diagram for water:



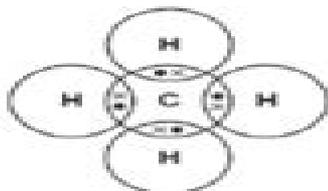
Dot and cross diagram for hydrogen chloride:



Dot and cross diagram for ammonia:



Dot and cross diagram for methane:



# Chapter 7 Ionic And Metallic Bonding Practice Problems Answers

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