



THIS WEEK'S HOT DEALS!

Red Grapes

99¢
Per Lb.

Blueberries

1.39
Per Pint

~~WAS 99¢~~

NOW 69¢
Per Lb.
Limit 4

FRESH FAMILY PACK
Chicken Drumsticks
4.5-lb. avg. weight

Mangoes

79¢
Each

89¢
Per 8-oz. Pkg.

Pennsylvania
whole white Mushrooms

California
Mandarins

2.59
Per 3-lb. Bag

1.89
Per 3-Pack

Multi-Colored Peppers

Available
Wednesday,
Feb. 20

8.99

FRESH
USDA CHOICE
Beef Pot. Roast Kit
2.7-lb.

Available
Wednesday,
Feb. 20

8.99
Per Lb.

FRESH
Norwegian
Atlantic
Salmon

Fresh.
We cook it here.

Available
Wednesday,
Feb. 20

1.29
Per Lb.

FRESH
Whole
Boneless
Pork Butt
Roast

Available
Wednesday,
Feb. 20

4.49
Per Lb.

FRESH FAMILY PACK
ORGANIC
GRASS-FED
85/15 Ground Beef
Sold in a 4-lb. pkg.
for \$17.99

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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^{-} + \text{Au} + \text{CN}^{-} + 2\text{e}^{-} \rightarrow 2\text{HCN} + 2\text{OH}^{-}$ $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ make electrons equal convert to basic solution $4\text{CN}^{-} + 2\text{Au} + 2\text{CN}^{-} + 2\text{e}^{-} \rightarrow 4\text{HCN} + 2\text{OH}^{-}$ multiplied by a factor of 2 $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ the final answer

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web aug 29 2023 the first step to balance any redox reaction is to separate the reaction into half reactions the substance being reduced will have electrons as reactants and the oxidized substance will have electrons as products usually all reactions are written as reduction reactions in half reaction tables

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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