

# UNIT 5 COORDINATION COMPOUNDS

## Multiple Choice Questions (MCQs):

- A purple compound 'M' has a formula  $\text{Co}(\text{NH}_3)_6\text{Cl}_2$ , on adding excess  $\text{AgNO}_3$  to one mole of 'M' produces 2 moles of  $\text{AgCl}$ . If secondary valence of Co is 6 then correct formula of the complex will be?
  - $[\text{CoCl}_2(\text{NH}_3)_6]$
  - $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$
  - $[\text{Co}(\text{NH}_3)_6\text{Cl}]\text{Cl}_2 \cdot \text{NH}_3$
  - $[\text{Co}(\text{NH}_3)_6\text{Cl}]\text{Cl}_2$
- Which among the following is not a double salt?
  - Mohr's Salt
  - Potash Alum
  - Carnallite
  - Potassium ferrocyanide.
- The correct order of the stoichiometries of  $\text{AgCl}$  formed when  $\text{AgNO}_3$  in excess is treated with the complexes:  $\text{CoCl}_2 \cdot 6\text{NH}_3$ ,  $\text{CoCl}_2 \cdot 5\text{NH}_3$ ,  $\text{CoCl}_2 \cdot 4\text{NH}_3$  respectively is (coordination valency is 6):
  - 3  $\text{AgCl}$ , 1  $\text{AgCl}$ , 2  $\text{AgCl}$
  - 3  $\text{AgCl}$ , 2  $\text{AgCl}$ , 1  $\text{AgCl}$
  - 2  $\text{AgCl}$ , 3  $\text{AgCl}$ , 1  $\text{AgCl}$
  - 1  $\text{AgCl}$ , 3  $\text{AgCl}$ , 2  $\text{AgCl}$
- Which among the following has highest molar conductivity?
  - $\text{K}_4[\text{Fe}(\text{CN})_6]$
  - $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$
  - $[\text{Cu}(\text{NH}_3)_4]\text{SO}_4$
  - $[\text{Ni}(\text{CO})_4]$
- The primary and secondary valencies of  $[\text{Fe}(\text{C}_2\text{O}_4)_3]^{3-}$  complexes are
  - 3 and 3
  - 3 and 6
  - 6 and 3
  - 0 and 3
- Coordination number of the central atom/ion is determined by
  - only by the number of pi bonds formed by the ligand with the central atom/ion.
  - both by the number of sigma and pi bonds formed by the ligand with the central atom/ion.
  - only by the number of sigma bonds formed by the ligand with the central atom/ion.
  - by the total number of electrons present in ligand.
- The correct IUPAC name of  $\text{K}_3[\text{Cr}(\text{C}_2\text{O}_4)_3]$  is
  - Potassiumtrioxalatochromium(III)
  - Potassiumtrioxalatochromate(III)
  - Potassiumtrisoxalatochromate(III)
  - Potassium oxalatochromate(III)
- $[\text{Pt}(\text{NH}_3)_4\text{Cl}_2] \text{Br}_2$  complex can show :
  - Hydrated as well as ionization isomerism
  - Ionization as well as geometrical isomerism
  - Linkage as well as geometrical isomerism
  - Ionization as well as optical isomerism
- Possible isomerism in complexes  $[\text{Co}(\text{NH}_3)_5(\text{NO}_2)]$  and  $[\text{Co}(\text{NH}_3)_5(\text{NO}_2)]\text{Cl}_2$  respectively are:
  - Linkage and optical
  - Geometrical and linkage
  - Optical and ionization
  - Linkage and geometrical
- In the reactions  $[\text{CoCl}_2(\text{NH}_3)_4]^{+} + \text{Cl}^{-} \rightarrow [\text{CoCl}_3(\text{NH}_3)_3] + \text{NH}_3$  two isomers of the product are obtained. The initial complex is
  - cis* isomer
  - trans* isomer
  - cis* or *trans* isomers
  - None of these

# Coordination Compounds Multiple Choice Questions

**Hussin A.Rothana**



**Coordination Compounds Multiple Choice Questions:**

The Enigmatic Realm of **Coordination Compounds Multiple Choice Questions**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Coordination Compounds Multiple Choice Questions** a literary masterpiece penned with a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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