

Stoichiometry Stumper #1

You are a forensic scientist. You are investigating a murder involving poison. The victim was poisoned with a compound called di-chloro benzene whose formula is $C_6H_4Cl_2$. Autopsy results show that the victim's body contained about 31 g of the poison, but the actual amount could have been slightly higher due to tissue absorption. The main suspect is his wife, Suzanne, who works as a chemistry professor at the local university. Records show that she purchased 15 g of benzene (C_6H_6) two days before the murder. Benzene is one of the compounds used to make the poison, but she claims she was using it to make ethyl benzene ($C_6H_5CH_3$), an innocuous compound, for use in her lab. She shows you the bottle of ethyl benzene she claims to have made. It contains 25 grams of ethyl benzene.

Is she telling the truth or did she have more nefarious motives? If you can show that it is possible to produce 25 g of ethyl benzene from 15 grams of benzene, then she was telling the truth. Otherwise, you will have caught her in a lie, which makes it likely she killed her husband with the poison. After extensive research in the literature, you find the two reactions related to this case.

Poisoned
To produce di-chloro benzene, the reaction is:



To produce ethyl benzene, the reaction is:



After balancing reactions, use stoichiometry to solve this case. Be sure to show all your work and explain whether the results show the wife to be innocent or a murderer.

Stoichiometry Stumper 1 Answer Forensic

TD Snyder



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