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 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement
 $4\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2$

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 $\text{NaOH} + \text{KNO}_3 \rightarrow \text{NaNO}_3 + \text{KOH}$ ch 2 o co 2 h 4 2 2 2o 2 fe 6 nabr 2 febr 6 na 3 caso mg
 $\text{OH} + \text{Ca(OH)}_2 + \text{MgSO}_4 \rightarrow 2\text{NH}_4\text{OH} + \text{H}_2\text{O} + \text{NH}_4\text{Br} + \text{PbO}_2 + \text{PbO}_2 + \text{Na}_2\text{CO}_3 + \text{Na}_2\text{O} + \text{CO}_2 + \text{C}_4\text{H}_8 + 6\text{O}_2 + 2\text{H}_2\text{O} + \text{HCl} + \text{NaOH} + \text{H}_2\text{O} + \text{NaCl}$

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