

### Student Exploration: Half-Life

**Prior Knowledge Questions** (Do these *before* starting the Lesson.)

1. What are some stable isotopes present? How do they not have either the property of an unstable isotope?

None. I have no prepping record.

2. If you had the opportunity to communicate on the topic of prepping isotopes, what are three things I should know?

Isotopes don't have to be used to make bombs, they can be used for medicine.

**Warm-Up**

Like an unstable isotope, the substance I **started with** went through a change in quantity. It had a **half-life** of 100 years. I had 1000 grams of it. After 100 years, I had 500 grams left. After 200 years, I had 250 grams left. After 300 years, I had 125 grams left. After 400 years, I had 62.5 grams left. After 500 years, I had 31.25 grams left. After 600 years, I had 15.625 grams left. After 700 years, I had 7.8125 grams left. After 800 years, I had 3.90625 grams left. After 900 years, I had 1.953125 grams left. After 1000 years, I had 0.9765625 grams left.

The half-life of a substance is the time it takes for half of the substance to decay. The half-life of a substance is 100 years. The substance I started with had a half-life of 100 years. The substance I started with had a half-life of 100 years. The substance I started with had a half-life of 100 years. The substance I started with had a half-life of 100 years.

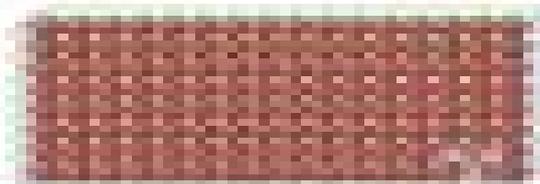


Figure 1: Exponential decay of a substance with a half-life of 100 years.

3. What do you know about half-life?

It's the time it takes for half of the substance to decay.

**Notes:** The following information will be used to create a **graph** of the **substance's** **decay** over time. The graph will show the amount of substance remaining over time. The graph will show the amount of substance remaining over time. The graph will show the amount of substance remaining over time. The graph will show the amount of substance remaining over time.

4. What do you know about the half-life of a substance?

It's the time it takes for half of the substance to decay.

5. In the graph, what is the half-life of the substance? How do you know?

The half-life is 100 years. It's the time it takes for half of the substance to decay.



# Half Life Gizmo Answer Key

**Norman Whitby**

A red circular graphic with a gradient, appearing as a semi-circle on the right side of the slide, partially overlapping the grey bar.

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