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endothermic
rh gt 0
b how much heat will be absorbed or released if 10 g of NO_2 is reacted with 10 g H_2O

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time

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 $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l})$ energy endothermic the transfer of heat from the surroundings to the system
energy $2\text{Hg}(\text{s}) + 2\text{Hg}(\text{l}) + \text{O}_2(\text{g})$

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