

One

Ways to Make the Most of Your **Credit Card Offers**



Credit Card Offers Update Login

R Bogdan



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 $2 \text{ } 3 \text{ ca} + \text{oh}_2 + 1 \text{ } \text{al}_2 \text{ so}_4 \rightarrow 3 \text{ } \text{caso}_4 + 2 \text{ al} + \text{oh}_3$ type of reaction double displacement
 $3 \text{ } 3 \text{ mg} + 1 \text{ } \text{fe}_2\text{o}_3 \rightarrow 2 \text{ } \text{fe} + 3 \text{ } \text{mgo}$ type of reaction single displacement
 $4 \text{ } 1 \text{ } \text{c}_2\text{h}_4 + 3 \text{ } \text{o}_2 \rightarrow 2 \text{ } \text{}$

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 $2 \text{ } 1 \text{ } \text{caco}_3 \rightarrow 1 \text{ } \text{cao} + 1 \text{ } \text{co}_2$ type of chemical reaction decomposition

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web balance the following equations and indicate the type of reaction taking place 1 $\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{HBr} + \text{Na}_3\text{PO}_4$ type of reaction 2 $\text{Ca} + \text{OH}^- \rightarrow \text{Al}_2\text{SO}_4 + 3 \text{CaSO}_4 + \text{Al} + \text{OH}^-$ type of reaction 3 $\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{MgO}$ type of reaction 4 $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ type of reaction 5 $\text{PbSO}_4 + \text{PbSO}_3 + \text{O}_2$ type of reaction 6 NH_3 12