

Colligative Properties

Colligative properties are characteristics of a solution that depend on the ratio of the number of solute particle to solvent particles.

Freezing
Point
Depression



Boiling
Point
Elevation



Osmotic
Pressure



Vapor
Pressure
Lowering



Colligative Properties Of A Solution

George H. Fried, George J. Hademenos

Colligative Properties Of A Solution:

Biophysics & Biophysical Chemistry D. Das, 1982 Chemical Principles Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding **Eckert Animal Physiology** David J. Randall, Warren Burggren, Kathleen French, Roger Eckert, 2002 This classic animal physiology text focuses on comparative examples that illustrate the general principles of physiology at all levels of organisation from molecular mechanisms to regulated physiological systems to whole organisms in their environment This textbook is an authoritative and complete guide to the field of animal physiology which uses a threefold approach to teaching The Comparative Approach emphasises basic mechanisms but allows patterns of physiological function in different species to demonstrate how evolution creates diversity This approach encourages students to appreciate the underlying principles that govern physiological systems The Experimental Emphasis helps students to understand the process of scientific discovery and shows how our knowledge of physiology continually increases and finally the Integrative Approach presents information about specific physiological systems at all levels of organisation from molecular interactions to interactions between an organism and its environment n included Chemistry Made Simple John T. Moore, Ed.D., 2010-04-21 See the world one molecule at a time Chemistry helps us understand not only the world around us but also our own bodies CHEMISTRY MADE SIMPLE makes it fun Each chapter has practice problems with complete solutions that reinforce learning A glossary of chemical terms the modern periodic table and detailed illustrations throughout make this the best introduction to one of the most studied of all sciences Topics covered include the Scientific Method the structure and properties of matter compounds laws of chemistry gases liquids and solids solutions electrochemistry the atmosphere biochemistry organic chemistry nuclear chemistry energy the environment Look for these Made Simple titles Accounting Made Simple Arithmetic Made Simple Astronomy Made Simple Biology Made Simple Bookkeeping Made Simple Business Letters Made Simple Earth Science Made Simple English Made Simple French Made Simple German Made Simple Ingles Hecho Facil Investing Made Simple Italian Made Simple Latin Made Simple Learning English Made Simple Mathematics Made Simple The Perfect Business Plan Made Simple Philosophy Made Simple Physics

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