

# ACS Biochemistry Final

pKa (chapter 3)

measure of the tendency of a group to give up a proton (acidity); tendency decreases tenfold as pKa increases by one unit

thioester bond (1)

compounds with functional group C-S-CoA-C (eg, in acetyl-CoA)

zwitterion (3)

dipolar ion with both positive and negative groups but overall neutral charge; can act as either an acid or base

Isoelectric point (pI) (chapter 3)

the characteristic pH at which the net electric charge is zero

SDS - Sodium Dodecyl Sulfate (chap 3)

detergent used to unfold proteins and give them uniform negative charge

SDS Page (3)

chromatography used to separate proteins based on mass. light proteins travel faster than heavier ones

PCR - Polymerase chain reaction (3)

copies DNA multiple times to increase sample size

Isoelectric focusing (chapt 3)

procedure used to determine the isoelectric pt (pI) of a protein. Protein migrates through gel until pH = pI (net charge = 0)

Two-Dimensional Electrophoresis (3)

combines isoelectric focusing and SDS electrophoresis; separates proteins by both molecular weight and pI

specific activity (3)

number of enzyme units per mg of total protein (a measure of enzyme purity)

activity (3)

total units of a certain enzyme in a solution

Peptide bonds (chapter 4)

C-N bond with double bond character due to resonance (C-N bond cannot rotate, and is planar)

Edman degradation (3)

used in the sequencing of polypeptides; labels and removes ONLY the amino-residue from a polypeptide. carried out in a machine called a sequenator

$\phi$  in peptide bonding (chapter 4)

angle around the  $\alpha$ -carbon - amide nitrogen bond

$\psi$  in peptide bonding (chapter 4)

angle around the  $\alpha$ -carbon - carbonyl carbon bond

Ramachandran Plot (4)

shows favorable  $\phi$ - $\psi$  angle combinations. 3 main "wells" for  $\alpha$ -helices,  $\beta$ -sheets, and left handed  $\alpha$ -helices

Levinthal's Paradox (4)

# Biochemistry Acs Exam Study Guide 2013

**Padhraic Smyth**



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## **Table of Contents Biochemistry Acs Exam Study Guide 2013**

1. Understanding the eBook Biochemistry Acs Exam Study Guide 2013
  - The Rise of Digital Reading Biochemistry Acs Exam Study Guide 2013
  - Advantages of eBooks Over Traditional Books
2. Identifying Biochemistry Acs Exam Study Guide 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Biochemistry Acs Exam Study Guide 2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from Biochemistry Acs Exam Study Guide 2013
  - Personalized Recommendations
  - Biochemistry Acs Exam Study Guide 2013 User Reviews and Ratings
  - Biochemistry Acs Exam Study Guide 2013 and Bestseller Lists

5. Accessing Biochemistry Acs Exam Study Guide 2013 Free and Paid eBooks
  - Biochemistry Acs Exam Study Guide 2013 Public Domain eBooks
  - Biochemistry Acs Exam Study Guide 2013 eBook Subscription Services
  - Biochemistry Acs Exam Study Guide 2013 Budget-Friendly Options
6. Navigating Biochemistry Acs Exam Study Guide 2013 eBook Formats
  - ePub, PDF, MOBI, and More
  - Biochemistry Acs Exam Study Guide 2013 Compatibility with Devices
  - Biochemistry Acs Exam Study Guide 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Biochemistry Acs Exam Study Guide 2013
  - Highlighting and Note-Taking Biochemistry Acs Exam Study Guide 2013
  - Interactive Elements Biochemistry Acs Exam Study Guide 2013
8. Staying Engaged with Biochemistry Acs Exam Study Guide 2013
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Biochemistry Acs Exam Study Guide 2013
9. Balancing eBooks and Physical Books Biochemistry Acs Exam Study Guide 2013
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Biochemistry Acs Exam Study Guide 2013
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Biochemistry Acs Exam Study Guide 2013
  - Setting Reading Goals Biochemistry Acs Exam Study Guide 2013
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Biochemistry Acs Exam Study Guide 2013
  - Fact-Checking eBook Content of Biochemistry Acs Exam Study Guide 2013
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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