



# Nfl Schedule This Week Customer Service

**Michael Rajnik**



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web atoms of the same element i e same atomic number  $Z$  that have different numbers of neutrons are called isotopes for example  ${}^{12}_6\text{C}$  of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1 of the carbon atoms have 7

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web atoms of the same element that differ in their numbers of neutrons are called isotopes many isotopes occur naturally usually one or two isotopes of an element are the most stable and common different isotopes of an element generally have the same physical and chemical properties because they have the same numbers of protons and electrons

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nuclear particles i e z z 1 we gots hydrogen z 2 we gots helium z 3 lithium z 47 we gots silver the nucleus can also contain neutrons massive nuclear particles of zero charge

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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