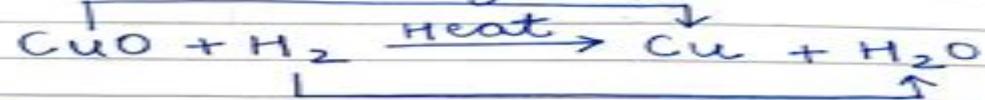


Reduction : Copper oxide is being reduced, CuO is an oxidising agent

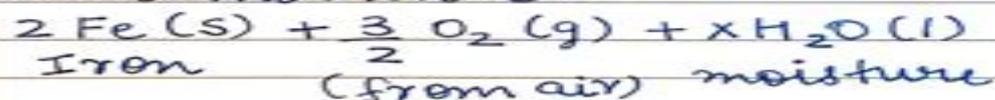


Oxidation : Hydrogen is being oxidised, H<sub>2</sub> is the reducing agent.

**OXIDATION**

**EFFECTS IN EVERYDAY LIFE**

**CORROSION**  
Slow conversion of metals into their undesirable compounds (sulphides, carbonates, oxides, etc.) by interaction with atmospheric gases and moisture



→ FeO<sub>3</sub> · xH<sub>2</sub>O (s)  
Hydrated ferric oxide (Rust)

**RANCIDITY**  
Oxidation of oils & fats in food resulting in bad taste and smell.

**METHODS OF PREVENTION**

- By adding Antioxidants ←
- Vacuum ~~vacuum~~ Packing ←
- Replacing air by Nitrogen ←
- Refrigeration of the Foodstuff ←

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**CL Gary**



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### **Table of Contents Cbse Class 10 Science Notes Golden**

1. Understanding the eBook Cbse Class 10 Science Notes Golden
  - The Rise of Digital Reading Cbse Class 10 Science Notes Golden
  - Advantages of eBooks Over Traditional Books
2. Identifying Cbse Class 10 Science Notes Golden
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Cbse Class 10 Science Notes Golden
  - User-Friendly Interface
4. Exploring eBook Recommendations from Cbse Class 10 Science Notes Golden
  - Personalized Recommendations
  - Cbse Class 10 Science Notes Golden User Reviews and Ratings
  - Cbse Class 10 Science Notes Golden and Bestseller Lists

5. Accessing Cbse Class 10 Science Notes Golden Free and Paid eBooks
  - Cbse Class 10 Science Notes Golden Public Domain eBooks
  - Cbse Class 10 Science Notes Golden eBook Subscription Services
  - Cbse Class 10 Science Notes Golden Budget-Friendly Options
6. Navigating Cbse Class 10 Science Notes Golden eBook Formats
  - ePub, PDF, MOBI, and More
  - Cbse Class 10 Science Notes Golden Compatibility with Devices
  - Cbse Class 10 Science Notes Golden Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Cbse Class 10 Science Notes Golden
  - Highlighting and Note-Taking Cbse Class 10 Science Notes Golden
  - Interactive Elements Cbse Class 10 Science Notes Golden
8. Staying Engaged with Cbse Class 10 Science Notes Golden
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Cbse Class 10 Science Notes Golden
9. Balancing eBooks and Physical Books Cbse Class 10 Science Notes Golden
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Cbse Class 10 Science Notes Golden
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Cbse Class 10 Science Notes Golden
  - Setting Reading Goals Cbse Class 10 Science Notes Golden
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Cbse Class 10 Science Notes Golden
  - Fact-Checking eBook Content of Cbse Class 10 Science Notes Golden
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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