

HOW THE GAME IS PLAYED

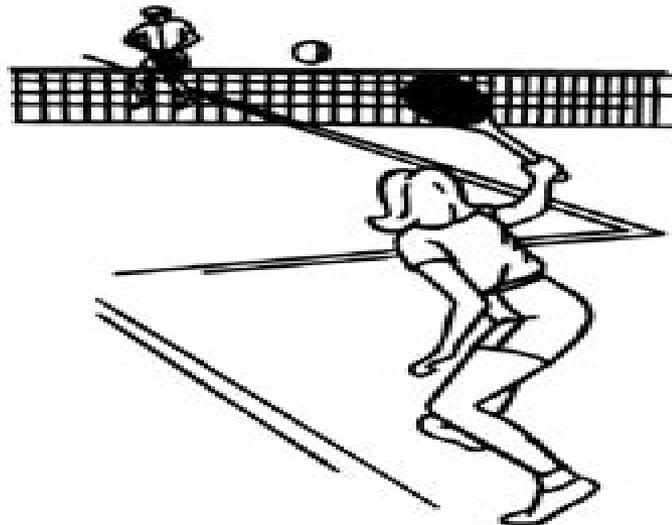
A tennis game always starts with the serve. The server stands behind the base line on the right side of the court. The ball is tossed into the air and hit diagonally across the net into the opponent's court. The opponent then has to hit the ball after the first bounce, and return it back across the net to the server.



A "fault" is committed when the server hits the net with the ball or when the ball is hit outside the service court. The second time a server tries to serve and fails, "a double fault" occurs. The opponent automatically gains a point after a double fault.

Faults can also occur when the server touches the base line or any other part of the court with his/her feet while serving.

A "let ball" is a serve that touches the net but lands in the correct place. This type of ball does not count; the server must serve again.



A ball that touches the net during play is called a "net ball." Net balls are legal and must be played.

Once the first point is made, the server then serves from behind the left base line. The server continues until the opponent wins the serve or until a fault occurs.

The first point in a tennis game is 15, the second is 30 and the third is 40. The final score (or fourth point) is the GAME POINT.

A player must score four points in order to win a game. In a tie game, both sides or players may score three points, or 40-40 (a "DEUCE"). If a player or side wins two more points after a deuce, the game is won.

The "AD" or ADVANTAGE is the first point scored after the deuce. When both players

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specifically designed to teach the students the skills of the game in a fun and interesting manner ultimately leading up to Team Handball This Sports lesson provides a teacher and student section with a variety of reading passages activities crossword word search and answer key to create a well rounded lesson plan **Modified Continental Handball for Inclusive Physical Education Classes** Claudette Prevost,British Columbia Teachers' Federation. Lesson Aids Service,1998 Grade level 3 4 5 6 7 8 9 10 11 12 k p e i s t *Modern Handball : Beginner to Expert* McGill University. Physical Education Department,Mezey, Andy,Graham I. Neil,1976

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the

syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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