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hack to get Windows XP running on these new Macs Enter Boot Camp a new system utility that helps you partition and install Windows XP on your Intel Mac Boot Camp does all the heavy lifting for you You won't need to open the Terminal and hack on system files or wave a chicken bone over your iMac to get XP running This free program makes it easy for anyone to turn their Mac into a dual boot Windows OS X machine Running Boot Camp guides you step by step through the entire installation process including upgrading your Mac's Firmware creating the Macintosh Drivers CD to make XP work properly with your Mac's hardware and using the Boot Camp Assistant to partition your hard drive and install Windows XP You'll also learn how to avoid common pitfalls such as previously partitioned drives and wrong disk permissions And finally you'll find out which Mac functions don't work in XP and which XP features backfire on a Mac With this invaluable guide at your side you'll finish configuring your dual boot Mac in as little as two hours avoiding numerous hazards and annoyances along the way

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web jun 14 2015 this list of electron configurations of elements contains all the elements in increasing order of atomic number to save room the configurations are in noble gas shorthand this means part of the electron configuration has been replaced with the element symbol of the noble gas symbol

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1s orbital and the third is placed in the 2s orbital giving an electron configuration of $1s^2 2s^1$

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are $he\ 2$ helium $1s\ 2$ $ne\ 10$ neon $1s\ 2\ 2s\ 2\ 2p\ 6$

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is $1s\ 2\ 2s\ 2\ 2p\ 6\ 3s\ 2\ 3p\ 6\ 4s\ 1$ or simply put as $4s\ 1$ q2 what is the total number of electrons that can be filled in s p and d subshells

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