

Kani's method for frame analysis

- Kani's method is an iterative technique for analyzing indeterminate structures, particularly useful for multi-storey rigid frames
- It simplifies complex calculations, focusing on rotational equilibrium at joints to determine moment distribution
- This method is efficient in handling asymmetrical buildings and structures with varying stiffness
- It calculates moments at the ends of each member, which helps in designing safer and more economical structures

Practical example: Structural engineering

- Kani's method is a vital tool in structural engineering for analyzing indeterminate structures, particularly frames
- This method simplifies the complex calculations needed to determine the bending moments and shear forces within a frame, which are crucial for assessing its stability and safety
- By iteratively adjusting moment distribution until equilibrium is achieved, engineers can ensure that structures like buildings and bridges can withstand applied loads
- The accuracy and efficiency of Kani's method make it a popular choice for designing safe and resilient structural frameworks



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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b

had a melting point range of 122.6–124.5 °C thus

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Nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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Melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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