

4. A solution contains 2.0 g of FeCl_3 in 100 mL of water. The molar mass of FeCl_3 is 162.5 g/mol. What is the molarity of the solution?
- 0.012 M
 - 0.024 M
 - 0.036 M
 - 0.048 M
5. A solution contains 2.0 g of FeCl_3 in 100 mL of water. The molar mass of FeCl_3 is 162.5 g/mol. What is the molarity of the solution?
- 0.012 M
 - 0.024 M
 - 0.036 M
 - 0.048 M
6. The molarity of a solution is 0.5 M. What is the mass of solute in 100 mL of the solution?
- 5.0 g
 - 10.0 g
 - 15.0 g
 - 20.0 g
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 - 0.036 M
 - 0.048 M

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