

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	56	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

Chapter 12 Basic Review Worksheet Chemistry Answers

W.H. Schmidt



Chapter 12 Basic Review Worksheet Chemistry Answers:

Science Insights ,1999 **Books in Print Supplement** ,2002 *Linne & Ringsrud's Clinical Laboratory Science - E-Book* Mary Louise Turgeon,2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts **Characterizing Pedagogical Flow** W.H. Schmidt,2005-12-17 Characterizing Pedagogical Flow presents conclusions from a multi disciplinary multi national research project blending quantitative and qualitative approaches through a discourse methodology The work produced portraits of mathematics and science education that were dramatically different for each of the countries involved France Japan Norway Spain Switzerland and the United States To explain these differences it is proposed that the interaction of curriculum and pedagogy is culturally unique and yields classroom learning experiences that are qualitatively different from country to country This idea has profound implications for how international education research is interpreted **Biology** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004 Holt Chemistry Ralph Thomas Myers,2004

Curriculum Review ,1986 **Programmed Learning and Individually Paced Instruction** Carl H. Hendershot,1973

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