

# MAIN IDEA

## Reading Passage

### Mrs. Thompson's Treasures

Read the passage 3 times.  
Answer the questions below. Underline your evidence.

Name \_\_\_\_\_

Main Idea & Details



### Ice Cream

When you think about different flavors of ice cream, you may imagine chocolate, vanilla, and strawberry. But have you ever heard of ice cream with lobster in it? A shop in Bar Harbor, Maine serves Lobster Ice Cream, which has real lobster pieces cooked, buttered, and folded into a buttery vanilla ice cream. A shop in New York tries out new one-of-a-kind ice cream flavors each day. Some of their creations include Garlic Ice Cream, Pizza Ice Cream, and Corn on the Cob Ice Cream. For Thanksgiving, you can have your turkey dinner and dessert all in one at another New York shop that makes Brown Turkey Ice Cream. If you are feeling really adventurous, you can try you to try Ramen Town, an amusement park in Tokyo, Japan. They serve Ramen Cow Tongue Ice Cream, and even Ramen.

What is the main idea of the story?



- 1) Ramen Town is in Tokyo, Japan.
- 2) Ice cream can come in many unique flavors.
- 3) You can get Garlic Ice Cream in New York.
- 4) Chocolate and vanilla are the most popular flavors.



# FREE

# Reading Comprehension Ideas

**Lei Huang**



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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below  
 $\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$

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web you may have to consult the solubility rules write the net ionic equation for the reaction of  $\text{BaCl}_2(\text{aq})$  and  $\text{Na}_2\text{SO}_4(\text{aq})$  you may have to consult the solubility rules write

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web choose the correct net ionic equation representing the reaction that occurs when solutions of potassium carbonate and copper i chloride are mixed a  $\text{K}_2\text{CO}_3$  aq  $2\text{CuCl}$  aq

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web the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in

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web the net ionic equation for equation 1 is given as equation 3  $\text{S}_2$  aq  $\text{Fe}_2$  aq  $\text{gt}$   $\text{fes}$  s eq 3 8 note that the phase indicator aq is optional on ionic species but is

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web  $\text{CuClO}_4$  2 is soluble all perchlorates are soluble while  $\text{PbSO}_4$  is insoluble sulfates are generally soluble except for  $\text{Pb}^{2+}$   $\text{Ba}^{2+}$   $\text{Hg}_2^{2+}$  and  $\text{Sr}^{2+}$  the net ionic equation is

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