

# Customer Service Response Templates: Best Practices



**Keep It Short and Friendly**



**Make It Suitable for Everyone**



**Personalize Messages Where Possible**



**Include a Call to Action**

# Act Practice How To Customer Service

**Raffaela Di Napoli**



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web we used mole fractions in gen chem 1 to describe gas mixtures where the partial pressure of a gas was its mole fraction time the total pressure  $p_a = x_a \cdot p_{total}$  mole fraction  $x_a = \frac{p_a}{p_{total}}$  mole fraction of substance  $a$  is often denoted by  $x_a$  and

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web nov 6 2023  $x_a$  mol  $a$  mol  $a$  mol  $b$  and  $x_b$  mol  $b$  mol  $a$  mol  $b$  note in the given mixture the sum of all the mole fractions present is equal to one  $x_a + x_b = 1$  when the mole fractions are multiplied by 100 they give the mole percentage mole fraction is a unitless and dimensionless expression image will be uploaded soon

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web solution total mass of the answer 100 g mass of benzene 30 g therefore mass of carbon tet 100 30 g benzene s molar mass  $C_6H_6$  128.17 g mol<sup>-1</sup> hence  $C_6H_6$  s number of moles  $\frac{30}{128.17}$  mol 0.234 mol molar mass of carbon tet  $CCl_4$  153.84 g mol<sup>-1</sup>

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web jan 30 2020 ans the molality of solution is 1 mol kg<sup>-1</sup> or 1 m example 0.2111 g of urea  $NH_2CONH_2$  was dissolved in 100 g of water calculate the molarity and molality of the solution given n 14 h 1 c 12 o 16 given mass of solute urea 11.11 g mass of solvent water 100 g 0.1 kg to find molarity of solution solution

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web solution moles of  $CH_3OH$  5.532017 mole moles of  $H_2O$  40.1822 moles therefore according to the equation mole fraction of  $CH_3OH$   $\frac{5.532017}{5.532017 + 40.1822}$  0.12073 to solve more examples on mole fraction formula and practice more question please visit byju s com

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web you may work out the mole fractions on your own problem 5 what is the mass of a sample of a 0.449 molal kbr that contains 2.92 kg of water 1 molality moles solute divided by kilograms solute 0.449 mol kg<sup>-1</sup> x 2.92 kg x 131.108 mol of kbr 2 moles times molar mass equals grams 131.108 mol 119.0023 g mol 156 g kbr

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web calculate the mass percent mole fraction molality and molarity of hexane  $C_6H_{14}$  0.660 g cm<sup>-3</sup> when it is mixed with heptane  $C_7H_{16}$  0.684 g cm<sup>-3</sup> in a 35.0550 ml ratio assume that the volumes are additive

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0.2885 its mole percent would be 28.85

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web since the mole fraction of the o<sub>2</sub> is 0.050 it contributes 0.050 or if you will 5.0% of the total pressure thus total pressure is 0.210 / 0.050 = 4.2 atm problem 9 a sample of 1.43 g of helium and an unweighed quantity of o<sub>2</sub> are mixed in a flask at room temperature

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web if you play all of your chords short or long your comping will sound predictable be sure to vary your comping rhythms between short long rhythms and rhythms that fall on off the beat lastly we come to rhythmic spontaneity the goal of comping chords is to be able to create interesting chordal rhythms on the spot

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web in jazz comping an abbreviation of accompaniment or possibly from the verb to complement is the chords rhythms and countermelodies that keyboard players piano or organ guitar players or drummers use to support a musician s improvised solo or melody lines it is also the action of accompanying and the left hand part of a solo pianist

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web apr 21 2020 this rhythmic pattern can begin on beat 1 it is often repeated over two bars but not always notice that the first chord is sustained and the second is not it also sounds great to displace this rhythmic pattern to later beats including the up beats there s no limit to which beat you could begin this pattern on

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usually a piano organ or guitar sometimes this would include a vibraphone gary burton could really comp or even a jazz harp  
it s also very common for a musician to comp for themselves

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