

Year 7 Chemistry Revision

Topic 1: Matter

1. The table below shows the melting points and boiling points of four substances, P, Q, R and S.

Substance	Melting Point (°C)	Boiling Point (°C)
P	0	100
Q	5	80
R	20	90
S	40	120

At which temperature(s) would all the substances be in the same state?

A: 3°C

B: 45°C

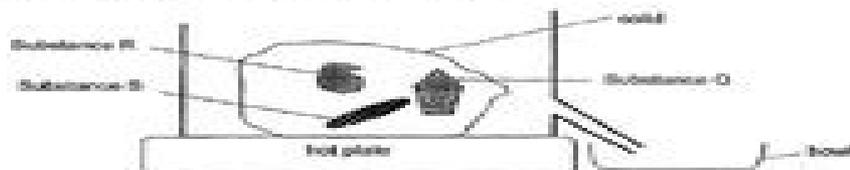
C: 100°C

D: 150°C

- a. B only
b. A and C only
c. B and D only
d. A, C and D only

[1]

2. Jack placed a solid made of substances Q, R and S in the set-up as shown below. Substances Q and R had higher melting points than substance S.



Before switching on the hot plate, he wanted to make sure that the liquid in the bowl contains only substance S. What temperature should he set for the hot plate?

Year 7 Chemistry Test Papers

J Elliott



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Table of Contents Year 7 Chemistry Test Papers

1. Understanding the eBook Year 7 Chemistry Test Papers
 - The Rise of Digital Reading Year 7 Chemistry Test Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Year 7 Chemistry Test Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Year 7 Chemistry Test Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Year 7 Chemistry Test Papers
 - Personalized Recommendations

- Year 7 Chemistry Test Papers User Reviews and Ratings
- Year 7 Chemistry Test Papers and Bestseller Lists
- 5. Accessing Year 7 Chemistry Test Papers Free and Paid eBooks
 - Year 7 Chemistry Test Papers Public Domain eBooks
 - Year 7 Chemistry Test Papers eBook Subscription Services
 - Year 7 Chemistry Test Papers Budget-Friendly Options
- 6. Navigating Year 7 Chemistry Test Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Year 7 Chemistry Test Papers Compatibility with Devices
 - Year 7 Chemistry Test Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Year 7 Chemistry Test Papers
 - Highlighting and Note-Taking Year 7 Chemistry Test Papers
 - Interactive Elements Year 7 Chemistry Test Papers
- 8. Staying Engaged with Year 7 Chemistry Test Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Year 7 Chemistry Test Papers
- 9. Balancing eBooks and Physical Books Year 7 Chemistry Test Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Year 7 Chemistry Test Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Year 7 Chemistry Test Papers
 - Setting Reading Goals Year 7 Chemistry Test Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Year 7 Chemistry Test Papers
 - Fact-Checking eBook Content of Year 7 Chemistry Test Papers

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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