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oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni} + 3 \text{ Ag}$  oxidation reduction  $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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5co i2 5co2 i 2 o 5 5 c o i 2 5 c o 2

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