

## Chemistry Lab Report: Identifying an Unknown Compound

In this lab, I identified an unknown compound by testing and comparing the experimental results of the compound with the known compounds. The unknown compound in this lab was a white, crystalline solid that melted at 100°C. I tested the compound for its solubility in water, its melting point, and its boiling point. I also tested the compound for its reactivity with acids and bases. The results of the tests are as follows: The compound is soluble in water, has a melting point of 100°C, and a boiling point of 100°C. It reacts with acids to form a colorless solution and with bases to form a colorless solution. The results of the tests indicate that the unknown compound is sodium chloride (NaCl).

The unknown compound and the known compounds were all tested with the same reagents. I used the same reagents for all tests. The results of the tests are as follows: The compound is soluble in water, has a melting point of 100°C, and a boiling point of 100°C. It reacts with acids to form a colorless solution and with bases to form a colorless solution. The results of the tests indicate that the unknown compound is sodium chloride (NaCl).

When I tested the unknown compound, I used the same reagents as the known compounds. The results of the tests are as follows: The compound is soluble in water, has a melting point of 100°C, and a boiling point of 100°C. It reacts with acids to form a colorless solution and with bases to form a colorless solution. The results of the tests indicate that the unknown compound is sodium chloride (NaCl).

I identified the unknown compound by testing and comparing the experimental results of the compound with the known compounds. The unknown compound in this lab was a white, crystalline solid that melted at 100°C. I tested the compound for its solubility in water, its melting point, and its boiling point. I also tested the compound for its reactivity with acids and bases. The results of the tests are as follows: The compound is soluble in water, has a melting point of 100°C, and a boiling point of 100°C. It reacts with acids to form a colorless solution and with bases to form a colorless solution. The results of the tests indicate that the unknown compound is sodium chloride (NaCl).

# Identifying Unknown Solutions Chemistry Lab Report

**LL Leslie**



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*Chemistry in the Laboratory* James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

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**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

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**Basic Laboratory Principles in General Chemistry**

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