



Classification of Crystalline Compounds

Identifying Polymers, Solids, Minerals, and Liquids

Rank of Crystalline Compounds

When crystalline polymers, minerals, and solids are heated, they undergo a phase transition and become amorphous. This transition is called the glass transition. The temperature at which this transition occurs is called the glass transition temperature (T_g). The T_g is a function of the polymer's molecular weight, crystallinity, and other factors. The T_g is also a function of the polymer's structure and the nature of the substituents on the polymer chain. The T_g is also a function of the polymer's processing history and the rate of cooling. The T_g is also a function of the polymer's environment and the presence of plasticizers.

When a polymer is heated above its T_g , it becomes a liquid. When a mineral is heated above its T_g , it becomes a liquid. When a solid is heated above its T_g , it becomes a liquid. When a liquid is heated above its T_g , it becomes a gas. When a gas is heated above its T_g , it becomes a plasma.

Questions: Rank of Crystalline Compounds

1. Which compound is not crystalline? Rank of Crystalline Compounds: (1) amorphous, (2) crystalline, (3) liquid, (4) solid, (5) gas, (6) plasma.
2. Which compound is not crystalline? Rank of Crystalline Compounds: (1) amorphous, (2) crystalline, (3) liquid, (4) solid, (5) gas, (6) plasma.
3. Which compound is not crystalline? Rank of Crystalline Compounds: (1) amorphous, (2) crystalline, (3) liquid, (4) solid, (5) gas, (6) plasma.
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10. Which compound is not crystalline? Rank of Crystalline Compounds: (1) amorphous, (2) crystalline, (3) liquid, (4) solid, (5) gas, (6) plasma.

Identifying Organic Compounds Lab Answers

Chao Zhang



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