

Solutions For Homework #1

Problem 1:

c is the speed of light, λ is the wavelength and f is the frequency. The equation $c = \lambda \cdot f$ can be used to calculate one of the three values, if the other two are given. The speed of light is defined to be exactly 299792458 m/s or $3 \cdot 10^8$ m/s. Thus, for a frequency of $f = 68.248 \cdot 10^6 \text{ s}^{-1} = 68.248 \text{ MHz}$,

$$\lambda = \frac{c}{f} = \frac{299792458 \text{ m/s}}{68248000 \text{ s}^{-1}} = \frac{3 \cdot 10^8 \text{ m/s}}{6.8248 \cdot 10^7 \text{ s}^{-1}} = 4.405 \text{ m} \approx 4.4 \text{ m}$$

and for a wavelength of $\lambda = 20 \text{ cm} = 0.20 \text{ m}$

$$f = \frac{c}{\lambda} = \frac{299792458 \text{ m/s}}{0.20 \text{ m}} = 1.5 \cdot 10^9 \text{ s}^{-1} = 1.5 \text{ GHz}$$

Problem 2:

The boiling temperature of water at sea level is 100°C. To convert this to degrees Kelvin, just add 273.15.

$$100^\circ\text{C} + 273.15 = 373.15 \text{ K} \approx 373 \text{ K}$$

Problem 3:

The formula for the wavelength of the energy radiation from a

$$\lambda_{\text{max}} = \frac{2898 \mu\text{m} \cdot \text{K}}{T}$$

where T is the black-body temperature. Using the temperatures for the surface of the sun ($T = 5778 \text{ K}$) and an effective value for the earth ($T = 255 \text{ K}$), we get
sun: 4.83 μm
earth: 11.36 μm

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Padhraic Smyth



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