

# Periodic Trends Notes

- The periodic table organizes all known elements according to their number of protons.
- The unique shape of the table reveals the ways in which the electrons are organized around the atom in energy levels.

Periodic Table of Elements

1A																	8A	
1	H																	He
2	Li	Be											B	C	N	O	F	Ne
3	Na	Mg	3B	4B	5B	6B	7B	8	9	10	11	12	13	14	15	16	17	18
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo		Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	*La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra	*Ac															

• Lanthanide Series

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd		Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

• Actinide Series

90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U											

Legend - click to find out more...

H - gas	Li - solid	Br - liquid	Tc - synthetic
<span style="background-color: #90EE90; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Non-Metals	<span style="background-color: #4682B4; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Transition Metals	<span style="background-color: #ADD8E6; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Rare Earth Metals	<span style="background-color: #FFFF00; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Halogens
<span style="background-color: #FFD700; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Alkali Metals	<span style="background-color: #7FFFD4; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Alkali Earth Metals	<span style="background-color: #800080; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Other Metals	<span style="background-color: #FF8C00; border: 1px solid black; display: inline-block; width: 15px; height: 15px;"></span> Inert Elements

# Notes Periodic Table And Periodic Trends I History

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student's understanding of these essential areas of chemistry. This new and updated edition of Periodicity and the s and p Block Elements provides a compelling and accessible introduction to key periodic trends found within the s and p blocks of the periodic table and includes coverage of the elements themselves as well as the compounds they form. Additional chapters focus on acidity and basicity as well as on structure. The final chapter is entirely new to the second edition and contains a critical examination of many theories, models, and approaches to the study of the ideas explored in the book. Digital formats and resources: The second edition is available for students and institutions to purchase in a variety of formats and is supported by online resources. The e-book offers a mobile experience and convenient access along with functionality, tools, navigation features, and links that offer extra learning support. [www.oxfordtextbooks.co.uk/ebooks](http://www.oxfordtextbooks.co.uk/ebooks). Online resources include multiple choice questions for students to check their understanding and for registered adopters, figures and tables from the book.

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