

Chemistry Chapter 5 Exam

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to
a. Mendeleev. c. Bohr.
b. Moseley. d. Ramsay.
- _____ 2. Mendeleev left spaces in his periodic table and predicted several elements and their
a. atomic numbers. c. properties.
b. colors. d. radioactivity.
- _____ 3. Mendeleev attempted to organize the chemical elements based on their
a. symbols. c. atomic numbers.
b. properties. d. electron configurations.
- _____ 4. Mendeleev noticed that properties of elements usually repeated at regular intervals when the elements were arranged in order of increasing
a. atomic number. c. reactivity.
b. density. d. atomic mass.
- _____ 5. Mendeleev is credited with developing the first successful
a. periodic table.
b. method for determining atomic number.
c. test for radioactivity.
d. use of X rays.
- _____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar
a. properties. c. densities.
b. atomic numbers. d. colors.
- _____ 7. In developing his periodic table, Mendeleev listed on cards each element's name, atomic mass, and
a. atomic number. c. isotopes.
b. electron configuration. d. properties.
- _____ 8. Mendeleev predicted that the spaces in his periodic table represented
a. isotopes. c. permanent gaps.
b. radioactive elements. d. undiscovered elements.
- _____ 9. Mendeleev's table was called periodic because the properties of the elements
a. showed no pattern.
b. occurred at repeated intervals called periods.
c. occurred at regular time intervals called periods.
d. were identical.
- _____ 10. The person whose work led to a periodic table based on increasing atomic number was
a. Moseley. c. Rutherford.
b. Mendeleev. d. Cannizzaro.
- _____ 11. Moseley's work led to the realization that elements with similar properties occurred at regular intervals when the elements were arranged in order of increasing
a. atomic mass. c. radioactivity.
b. density. d. atomic number.
- _____ 12. Who used his experimental evidence to determine the order of the elements according to atomic number?
a. Meyer c. Stas

Modern Chemistry Chapter 5 Section 1 Answers

Halbert Powers Gillette



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Modern Biopharmaceuticals, 4 Volume Set** Jörg Knäblein, 2005-10-28 The biopharmaceutical market has come along way since 1982 when the first biopharmaceutical product recombinant human insulin was launched Over 120 such products are currently being marketed around the world including nine blockbuster drugs The global market for biopharmaceuticals which is currently valued at US 41 billion has been growing at an impressive compound annual growth rate of 21% over the previous five years With over one third of all pipe line products in active development are biopharmaceuticals this segment is set to continue outperforming the total pharmaceutical market and could easily reach US 100 billion by the end of this decade **Understanding Pathophysiology - ANZ adaptation** Judy Craft, Christopher Gordon, Sue E. Huether, Kathryn L. McCance, Valentina L. Brashers, 2018-09-19 NEW chapter on diabetes to highlight the prevalence of the disease in Australia and New Zealand Expanded obesity chapter to reflect the chronic health complications and comorbidities New concept maps designed to stand out and pull together key chapter concepts and processes Updated Focus on Learning Case Studies and Chapter Review Questions Now includes an eBook with all print purchases *The Independent*, 1914 **The Well-Trained Mind** Susan Wise Bauer, Jessie Wise, 2024-04-02 The classic handbook on home schooling updated for a new generation of parents and students Is your child getting lost in the system becoming bored losing his or her natural eagerness to learn If so it may be time to take charge of your child s education by doing it yourself The Well Trained Mind will instruct you step by step on how to give your child an academically rigorous comprehensive education from preschool through high school one that will train him or her to read to think to understand to be well rounded and curious about learning Veteran home educator Susan Wise Bauer outlines the classical pattern of education called the trivium which organizes learning around the maturing capacity of the child s mind and comprises three stages the elementary school grammar stage when the building blocks of information are absorbed through memorization and rules the middle school logic stage in which the student begins to think more analytically and the high school rhetoric stage where the student learns to write and speak with force and originality Using this theory as your model you ll be able to instruct your child whether full time or as a supplement to classroom education in all levels of reading writing history geography mathematics science foreign languages rhetoric logic art and music regardless of your own aptitude in those subjects A new optional Resource Recommendations Portal provides subscribers with curated lists of the best curricula for every grade level and learning style Thousands of parents have already used the methods described in The Well Trained Mind to create a truly superior education for children in their care You do have control over what and how your child learns The Well Trained Mind will give you the tools you ll need to teach your child with confidence and success **The Open Shelf**, 1915 **The Chautauquan**, 1885 **Resources in Education**, 1990 Holt Chemistry Ralph Thomas Myers, 2004 *Understanding Pathophysiology Australia and New Zealand Edition* Judy

Craft, Christopher Gordon, Sue E. Huether, Kathryn L. McCance, Valentina L. Brashers, 2022-10-15 Understanding Pathophysiology Australia and New Zealand Edition *The American Magazine*, 1913 **The Independent** William Livingston, 1913 The Baltimore Underwriter, 1910 **The Bookseller**, 1904 **Engineering and Contracting** Halbert Powers Gillette, 1930 **Structure Elucidation by Modern NMR** H. Duddeck, W. Dietrich, 2012-12-06 For several years we have been organizing seminars and workshops on the application of modern one and two dimensional NMR methods at the faculty of chemistry in the Ruhr University Bochum FRG and elsewhere addressing researchers and graduate students who work in the field of organic and natural products chemistry In 1987 we wrote a workbook *Strukturaufklärung mit moderner NMR Spektroskopie* Steinkopff Darmstadt FRG 1988 in German language based on our experience in these courses Many of the exercises described therein have been used in such courses and some of them have been shaped by the participants to a great extent The response of readers and discussions with colleagues from many countries encouraged us to produce an English translation in order to make the book accessible to a wider audience Moreover the content has been increased from 20 exercise examples in the German to 23 in the English version This book could not have been written in the present form without the help of a number of colleagues and therefore we acknowledge gratefully the generous supply of samples from and useful discussions with B Abegaz Addis Ababa Ethiopia U H Brinker Bingham New York USA E *The Journal of Education*, 1895 **Bookseller and the Stationery Trades' Journal**, 1883 **American Illustrated Magazine**, 1913 **Engineering and Mining Journal**, 1903

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