

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double arrow pointing towards the right. The substance(s) on the left side of the reaction is/are
 - reactant(s)**
 - product(s)
 - solvent
- An ionic chemical equation is represented by adding the
 - charges**
 - charges of the particles
 - number of particles
- An ionic equation that is written as aq indicates the substance is
 - solid
 - dissolved in water**
 - liquid
 - volatile
- The activity for precipitation reactions are called double-replacement
 - double-replacement reactions for the activity half-reactions
 - precipitation reactions
 - redox reactions
 - precipitation reactions for the activity half-reactions**
- The coefficients are chemical equations
 - indicate the number of moles of each substance that react**
 - show the number of grams of each substance that react
 - show the number of molecules of the substances
 - show the relative proportions of each atom
- If you're to balance the chemical $Fe + O_2 \rightarrow Fe_2O_3$, then
 - Fe will replace some of the oxygen**
 - Fe will replace some of the oxygen
 - Fe will replace some of the oxygen
 - Fe will replace some of the oxygen

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