

Name _____

Hour _____

Stoichiometry Quiz

Answer all of the stoichiometry questions below using dimensional analysis, the technique learned in class. Be sure to show all of your work in order to receive partial credit. Make sure you circle your final answer.

1. Use the equation to answer the questions below: $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

- a. How many grams of HCl are in 2.56 moles HCl?

- b. How many moles of NaOH are in 45.26 grams NaOH?

2. Use the equation to answer the questions below: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

- a. How many moles of oxygen are required to produce 2.65 moles H_2O ?

- b. How many moles of methane (CH_4) are required to make 3 moles CO_2 ?

Chapter 9 Stoichiometry Quiz Answers

Clifford Lane



Chapter 9 Stoichiometry Quiz Answers:

Decoding **Chapter 9 Stoichiometry Quiz Answers**: Revealing the Captivating Potential of Verbal Expression

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