

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 9

## STUDY GUIDE FOR CONTENT MASTERY

## Covalent Bonding

### Section 11 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

|               |          |            |            |        |
|---------------|----------|------------|------------|--------|
| covalent bond | molecule | sigma bond | exothermic | p bond |
|---------------|----------|------------|------------|--------|

When sharing electron pairs, the attachment between atoms that results is called a(n) **covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **p bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is a water molecule a(n) **covalent bond**, **sigma bond**, **exothermic**, or **p bond**?
  - a single atom
  - a molecule containing three atoms
  - a molecule containing two atoms
  - a molecule containing four atoms
- How many electrons are shared in a double covalent bond?
  - one
  - two
  - three
  - four
- Bond length is the distance between
  - two molecules of the same substance
  - the electron in two attached atoms
  - the nuclei of two attached atoms
  - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
  - As the electron bond, the stronger the bond.
  - As the electron bond, the weaker the bond.
  - As the electron bond, the longer the electron is a.
  - As the electron bond, the lower the bond dissociation energy.

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## CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

### Section 12 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write *true* or *false*.

- false** 1. Binary molecular compounds are generally composed of a metal and a nonmetal.
- false** 2. The second element in the formula of a binary compound is named using the suffix *-ite*.
- false** 3. The prefix *mono-* indicates three atoms.
- true** 4. The prefix *hepta-* indicates six atoms.
- true** 5. In naming the first element in a formula, the prefix *mono-* is not used.
- true** 6. For binary acids, the hydrogen part of the compound is named using the prefix *hydro-*.
- false** 7. An oxyacid contains only two elements.
- false** 8. If the name of the cation of an oxyacid ends in *-ite*, the acid name contains the suffix *-ate*.

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

## Column A

- c.**  $\text{N}_2\text{O}$
- f.**  $\text{NO}_2\text{CO}$
- g.**  $\text{H}_2\text{SO}_4$
- h.**  $\text{H}_2\text{SO}_3$
- i.**  $\text{HNO}_3$
- j.**  $\text{HNO}_2$
- k.**  $\text{HNO}$
- l.**  $\text{HNO}$

## Column B

- a.** hydrobromic acid
- b.** nitrogen trioxide
- c.** carbon monoxide
- d.** nitrous acid
- e.** nitrous
- f.** nitric acid
- g.** carbonic acid
- h.** bromic acid
- i.** carbon dioxide

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Study Guide for Content Mastery

# Chapter 7 Study Guide For Content Mastery Answer Key

**Ying-Ying Zheng**



**Chapter 7 Study Guide For Content Mastery Answer Key:**

The book delves into Chapter 7 Study Guide For Content Mastery Answer Key. Chapter 7 Study Guide For Content Mastery Answer Key is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chapter 7 Study Guide For Content Mastery Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
    - Chapter 1: Introduction to Chapter 7 Study Guide For Content Mastery Answer Key
    - Chapter 2: Essential Elements of Chapter 7 Study Guide For Content Mastery Answer Key
    - Chapter 3: Chapter 7 Study Guide For Content Mastery Answer Key in Everyday Life
    - Chapter 4: Chapter 7 Study Guide For Content Mastery Answer Key in Specific Contexts
    - Chapter 5: Conclusion
  2. In chapter 1, the author will provide an overview of Chapter 7 Study Guide For Content Mastery Answer Key. The first chapter will explore what Chapter 7 Study Guide For Content Mastery Answer Key is, why Chapter 7 Study Guide For Content Mastery Answer Key is vital, and how to effectively learn about Chapter 7 Study Guide For Content Mastery Answer Key.
  3. In chapter 2, the author will delve into the foundational concepts of Chapter 7 Study Guide For Content Mastery Answer Key. This chapter will elucidate the essential principles that need to be understood to grasp Chapter 7 Study Guide For Content Mastery Answer Key in its entirety.
  4. In chapter 3, this book will examine the practical applications of Chapter 7 Study Guide For Content Mastery Answer Key in daily life. This chapter will showcase real-world examples of how Chapter 7 Study Guide For Content Mastery Answer Key can be effectively utilized in everyday scenarios.
  5. In chapter 4, this book will scrutinize the relevance of Chapter 7 Study Guide For Content Mastery Answer Key in specific contexts. This chapter will explore how Chapter 7 Study Guide For Content Mastery Answer Key is applied in specialized fields, such as education, business, and technology.
  6. In chapter 5, the author will draw a conclusion about Chapter 7 Study Guide For Content Mastery Answer Key. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 7 Study Guide For Content Mastery Answer Key.

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