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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains  
more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of  
monatomic ions include na fe 3 cl and many others

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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic  
formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen  
atoms with an overall charge of 1

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web polyatomic ions have defined formulas names and charges that cannot be modified in any way table 3 4 1 3 4 1 lists the ion names and ion formulas of the most common polyatomic ions for example no<sub>3</sub> no<sub>3</sub> is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion no<sub>3</sub> contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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