

# Calculating Tips

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Round off the bill.  
To Nearest 10

10% Tip

15% Tip

20% Tip



Round off the bill.  
To Nearest 10

10% Tip

15% Tip

20% Tip



Round off the bill.  
To Nearest 10

10% Tip

15% Tip

20% Tip



Round off the bill.  
To Nearest dollar

10% Tip

15% Tip

20% Tip



Round off the bill.  
To Nearest 10

10% Tip

15% Tip

20% Tip



Round off the bill.  
To Nearest 10

10% Tip

15% Tip

20% Tip



Round off the bill.  
To Nearest 10

10% Tip

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# Math Worksheet Tips Setup

**DJ Losen**



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below the formula of each of the reactants and products iii write down the relative masses of the reactants and products with  
the help of formula below the respective formula these shall be the theoretical amounts of reactants and

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conversion factor as another step in a calculation to make a mole mass calculation where we start with a given number of  
moles of a substance and calculate the mass of another substance involved in the chemical equation or vice

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moles of  $O_2$  in the jar then calculate the mass of the gas in the jar  
 d e f 12 mol  $O_2$  2 mol  $O_2$  6 mol  $O_2$  mass 384 g mass 64 g  
 mass 192 g 3 jars g h and i contain ammonia gas  $NH_3$

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