

13 These calculations have used $L = 6.02 \times 10^{23}$

- (a) 7.22×10^{22} (b) 3.01×10^{23}
 (c) 1.20×10^{23}

14 0.53 mol H

15 0.250 mol

- 16 (a) 262.87 g mol⁻¹ (b) 176.14 g mol⁻¹
 (c) 184.10 g mol⁻¹ (d) 248.22 g mol⁻¹

17 189.1 g

18 1.5 mol

19 0.0074 mol Cl⁻

20 1.83×10^{23} C atoms

21 171 g (integer value because no calculator)

22 10.0 g H₂O

23 2.0 mol N₂ > 3.0 mol NH₃ > 25.0 mol H₂ >
 1.0 mol N₂H₄

- 24 (a) CH (b) CH₂O
 (c) C₂H₂O₂ (d) C₂H₄
 (e) C₂H₄ (f) CH₂O

25 Na₂S₂O₅

26 CoSO₄·7H₂O

27 C₁₀H₁₆N

28 NH₃

29 5.94 L

30 CoS

31 empirical formula CH; molecular formula C₇H₈

32 empirical formula H₃PO₃; molecular formula H₄P₂O₆

33 C₁₂H₁₂N₂P₂O₁₂ for both empirical and molecular formulas

34 C₇H₈O

35 Let y = mass of chalk in grams.

$$\text{moles of chalk used} = \frac{\text{mass used}}{M_r(\text{CaCO}_3)} \times$$

$$y \frac{1 \text{ g}}{100.09 \text{ g mol}^{-1}}$$

This is the same as the number of moles of carbon atoms used.

Therefore the number of carbon atoms used = moles of chalk \times (6.02 \times 10²³ mol⁻¹) =

$$\frac{6.02 \times 10^{23} y}{100.09}$$

- 36 (a) 2.50 mol (b) 5.63 mol
 (c) 665.5 g

37 (a) $2\text{C}_2\text{H}_6 + 13\text{O}_2 \rightarrow 8\text{CO}_2 + 10\text{H}_2\text{O}$
 (b) 1.59 g

38 4.355 kg

- 39 (a) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
 (b) 92.8%

(c) CaCO₃ is the only source of CO₂; all the CaCO₃ undergoes complete decomposition; all CO₂ released is captured; heating does not cause any change in the mass of the other minerals present.

- 40 (a) 65.2 g (b) 1.3 g H₂

41 5.23 g C₇H₈Cl₂

42 253.2 g theoretical CaSO₄; 77.9%

43 3.16 g ester

44 107 g of C₇H₈ needed

- 45 (a) 2.40 mol (b) 0.0110 mol
 (c) 44 mol

- 46 (a) 35.65 dm³ (b) 5.7 dm³

47 0.652 dm³

48 0.138 mol Br₂ and 0.156 mol Cl₂, so more molecules of Cl₂

49 0.113 dm³

50 0.28 dm³

51 90 kPa

52 16 °C

53 3.0 dm³

54 2.8 dm³

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Lauren Gardner



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 - The Rise of Digital Reading Pearson Chemistry Workbook Answers Chapter 14
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Workbook Answers Chapter 14
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Workbook Answers Chapter 14
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Workbook Answers Chapter 14
 - Personalized Recommendations
 - Pearson Chemistry Workbook Answers Chapter 14 User Reviews and Ratings
 - Pearson Chemistry Workbook Answers Chapter 14 and Bestseller Lists

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 - Highlighting and Note-Taking Pearson Chemistry Workbook Answers Chapter 14
 - Interactive Elements Pearson Chemistry Workbook Answers Chapter 14
8. Staying Engaged with Pearson Chemistry Workbook Answers Chapter 14
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Workbook Answers Chapter 14
9. Balancing eBooks and Physical Books Pearson Chemistry Workbook Answers Chapter 14
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Workbook Answers Chapter 14
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pearson Chemistry Workbook Answers Chapter 14
 - Setting Reading Goals Pearson Chemistry Workbook Answers Chapter 14
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Chemistry Workbook Answers Chapter 14
 - Fact-Checking eBook Content of Pearson Chemistry Workbook Answers Chapter 14
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web jedes dieser genres bietet unterschiedliche möglichkeiten die themen und herausforderungen der adoleszenz zu erkunden so kann zum beispiel ein realistischer roman das alltägliche leben von jugendlichen darstellen während ein fantasy roman die gleichen themen in einem völlig anderen kontext behandeln kann unabhängig vom

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web adoleszenz und teenage2 bilden sich heraus so stellt sich in der literaturwissenschaft unter anderem die frage wie diese lebensphasen in literarischen texten aufgegriffen und konstruiert werden aber auch wie im umkehrschluss literarische texte für diese phase des lebens konzipiert werden