

## Review for Ch. 8 Test

**8.1 Describing Chemical Reactions**

- know how to write word equations (i.e. you have be able to name compounds, including acids, ionic, and covalent compounds from Ch. 7)
- know how to write a formula equation
- What 2 things do coefficients represent in an equation? **Relative number of moles of a substance ; relative number of molecules needed for a reaction to occur**
- know common symbols used in a reaction
- know 4 evidences of a chemical reaction:
  1. **energy in the form of heat or light**
  2. **production of a gas**
  3. **change in color**
  4. **formation of a precipitate**
- know the Law of conservation of matter; **matter can neither be created nor destroyed in a chemical reaction (which means, mass of reactants must equal the mass of products)**
- know what a chemical equation represents: **relative amounts of reactants and products**
- know how to balance chemical equations (combustion reactions are particularly tricky, practice!)

1. Complete the equation and write a word equation for the following reaction:  $\text{ZnO} + \text{FeSO}_4 \rightleftharpoons$



**Zinc oxide reacts with iron (II) sulfate to yield zinc sulfate and iron (II) oxide**

2. Write a formula equation for the reaction of a solution of AgCl with potassium metal:



3. What elements are always diatomic gases at room temperature?

**All of the halogens (Group 17) ,  $\text{H}_2$ ,  $\text{N}_2$ ,  $\text{O}_2$**

4. Can you balance these reactions:



# Describing Chemical Reactions Section Review Answer Key

**N Noddings**



## **Describing Chemical Reactions Section Review Answer Key:**

Holt Chemistry Ralph Thomas Myers,2004     Holt Chemistry Holt Rinehart & Winston,2003-01-24     *Sierra Nevada Forest Plan Amendment: Chapter 3, part 4* ,2001     **Radioactive Waste Management** ,1990-07     **Designs for Science Literacy** American Association for the Advancement of Science,2001-03-22 The call for science curriculum reform has been made over and over again for much of the twentieth century Arguments have been made that the content of the curriculum is not appropriate for meeting the individual and social needs of people living in the modern world that the curriculum has become overstuffed with topics and does not serve students especially well and above all that the curriculum does not generate the student learning it is expected to produce The latest volume in a continuing series of publications from the AAAS designed to reform science education *Designs for Science Literacy* presupposes that curriculum reform must be considerably more extensive and fundamental than the tinkering with individual courses and subjects that has been going on for decades *Designs* deals with the critical issues involved in assembling sound instructional materials into a new coherent K 12 whole The book pays special attention to the need to link science oriented studies to the arts and humanities and also proposes how to align the curriculum with an established set of learning goals while preserving the American tradition of local responsibility for the curriculum itself If fundamental curriculum reform is ever to occur a new process for creating alternatives will have to be developed *Designs for Science Literacy* provides the groundwork for such a process     Study Guide for Anatomy & Physiology - E-Book Linda Swisher, Kevin T. Patton, Gary A. Thibodeau, 2012-05-04 Each chapter will be updated to include revised content in the core textbook Addition of new Case Studies for each chapter     **Glencoe Science** ,2002     **Lehninger Biochemistry: Core Concepts and Applications** Michael Cox, Aaron Hoskins, Alain Viel, Judith Simcox, 2025-02-04 Dive into the world of Biochemistry with Achieve for Lehninger Biochemistry Core Concepts and Applications your gateway to mastering this complex field     *Chemistry* Neil D. Jespersen, Alison Hyslop, 2021-11-02 *Chemistry The Molecular Nature of Matter* 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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